



# Aquarius

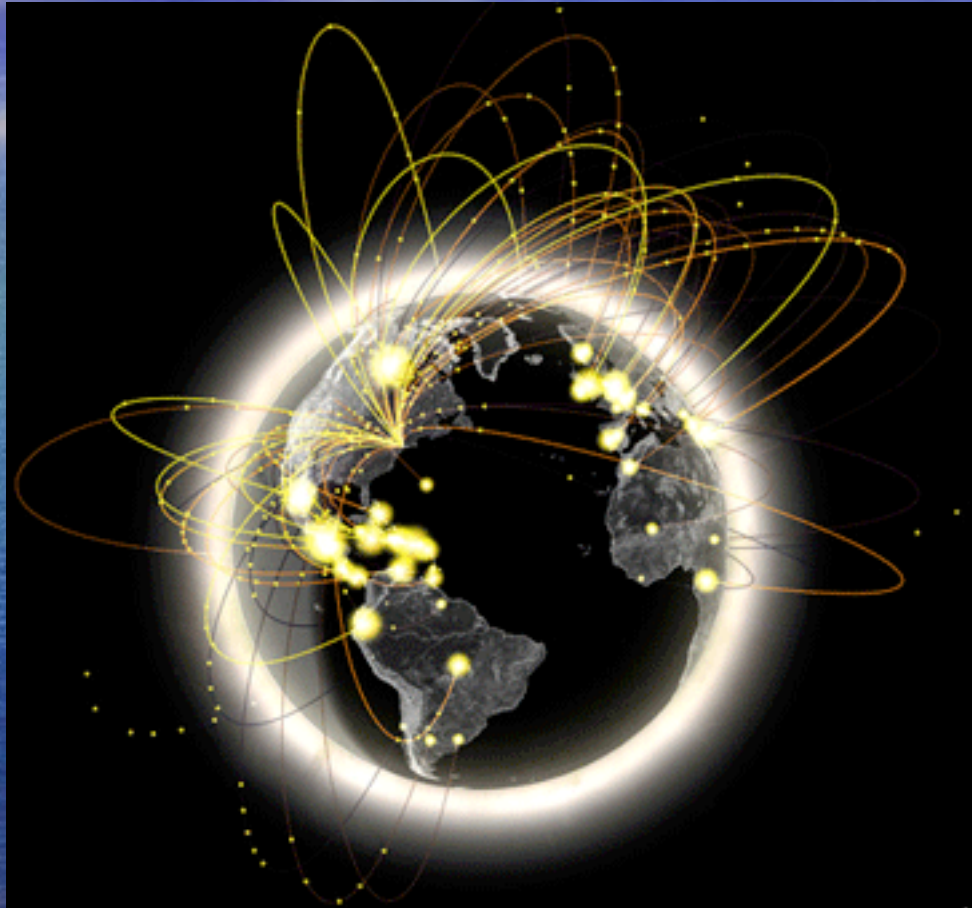
## Sea Surface Salinity from Space Teacher's Workshop

Federation of Earth Science  
Information Partners (ESIP)

Gregg Foti  
Physical Oceanography, Distributed Active Archive Center  
PO.DAAC

July 8<sup>th</sup>, 2009







# Overview

- Why is ocean remote sensing important?
- In Situ (place) measurements
- Satellite Instruments, including Aquarius
- Educational Tools & Activities
- PO.DAAC Products

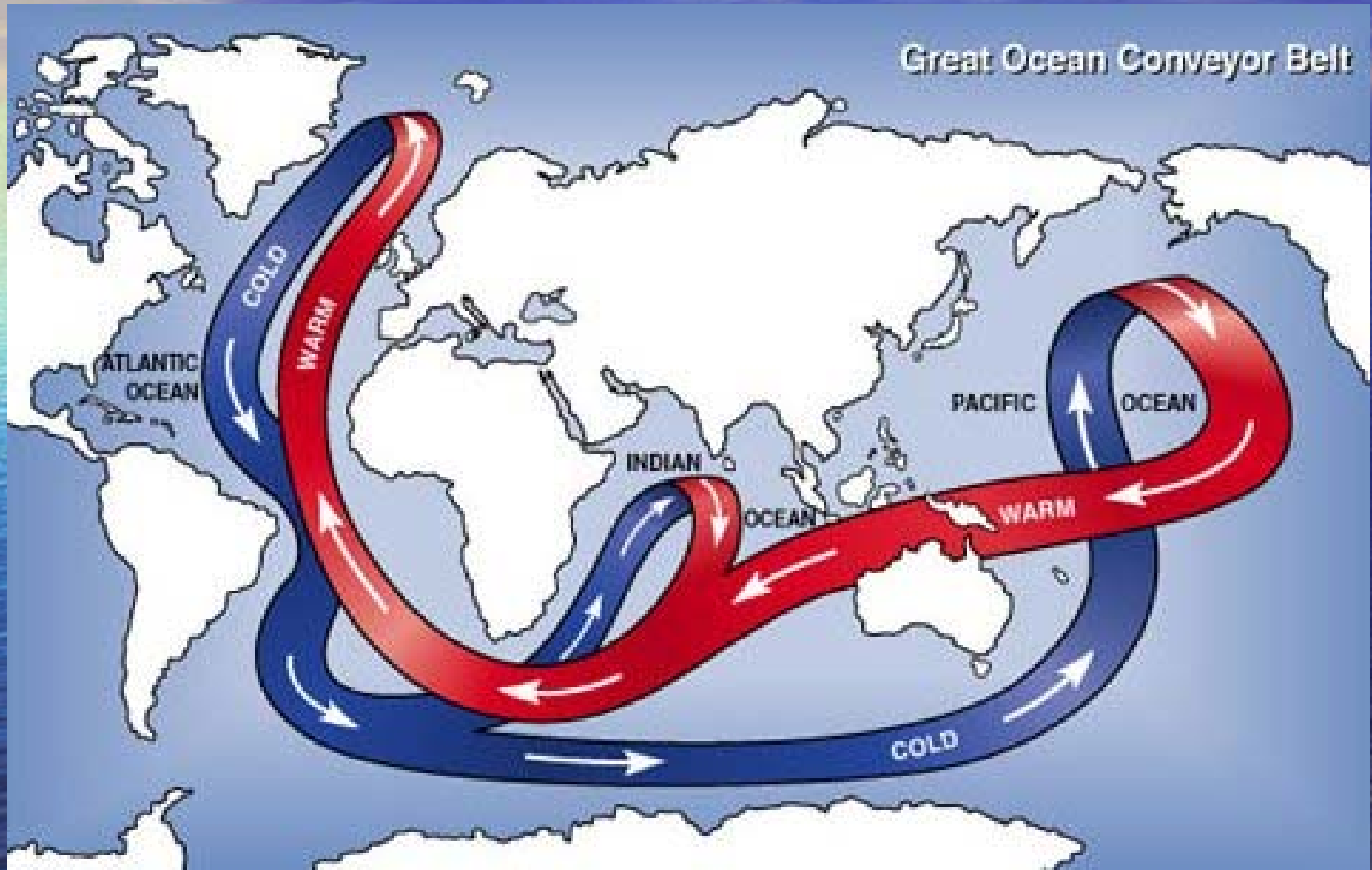
# Water cycle Ocean vs. Land

- ~70% of earth is covered by water
- Volume of water in oceans ~  $1.4 \times 10^9 \text{ km}^3$ 
  - 24 times as much as glaciers and lakes
  - 100,000 times as much as the atmosphere holds
- Heat capacity is 1100 times higher than atmosphere

# Global Water Cycle Fluxes

- Rainfall over oceans ~ 12 Sverdrups (Sv, 1 Sv = 1 million  $\text{m}^3/\text{s}$ )
- Rainfall over land ~ 3 Sv
- Evaporation over ocean ~ 13 Sv
- Evaporation over land ~ 2 Sv
- Terrestrial Runoff ~ 1 Sv
  
- Water Cycle is dominated by ocean - atmosphere fluxes!

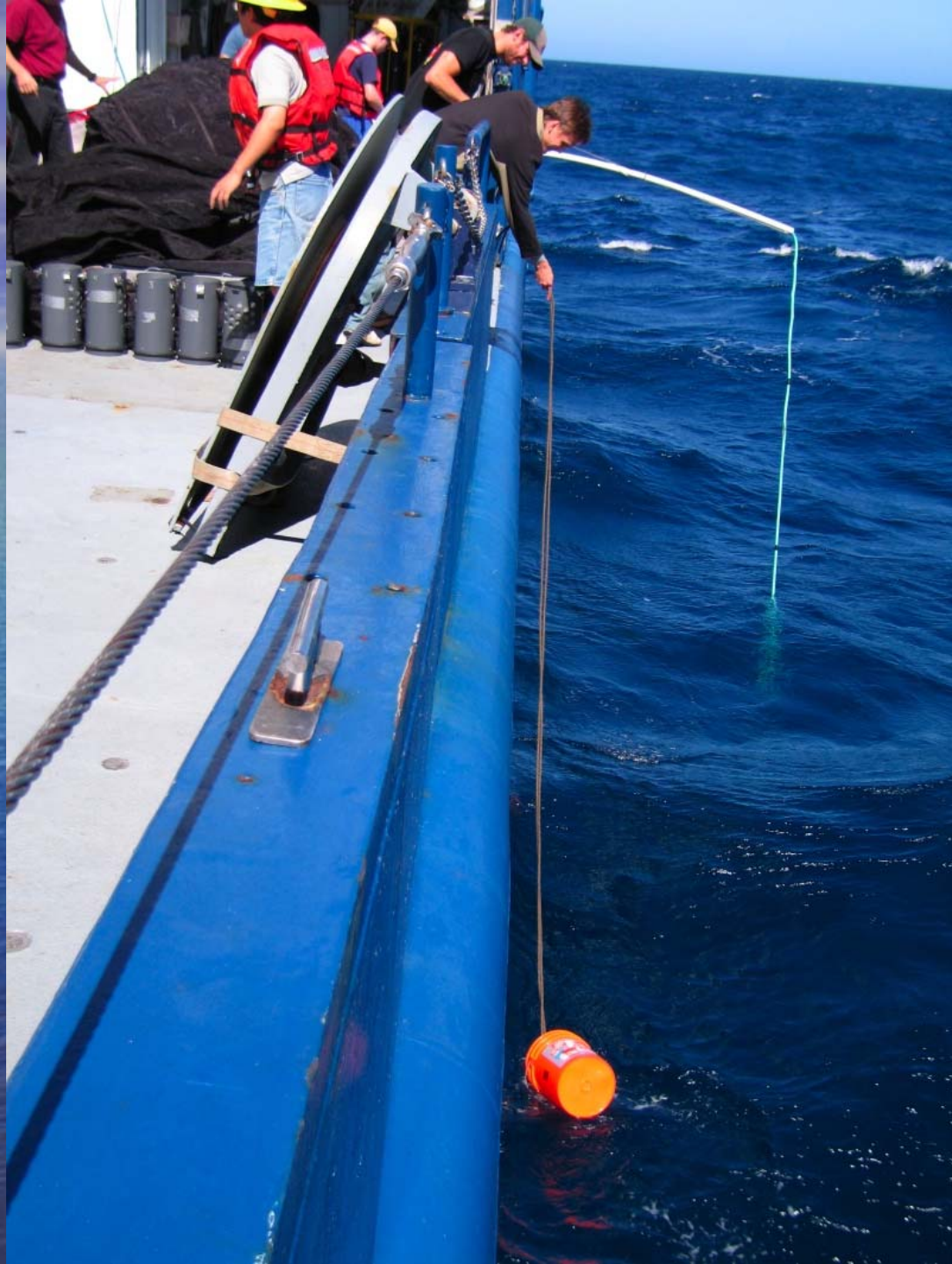
# Conveyor Belt

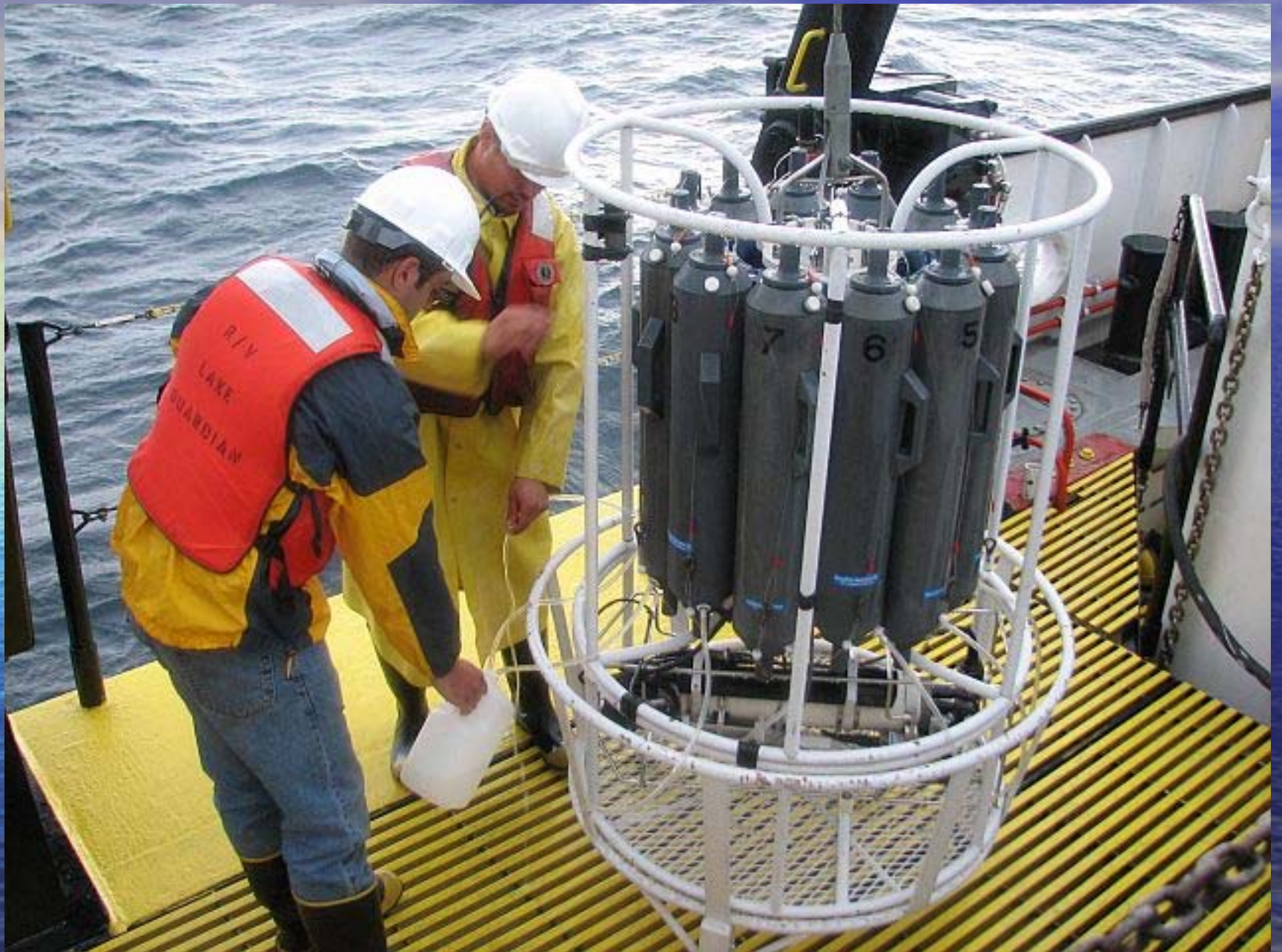




# In Situ Measurements







# Salinometer

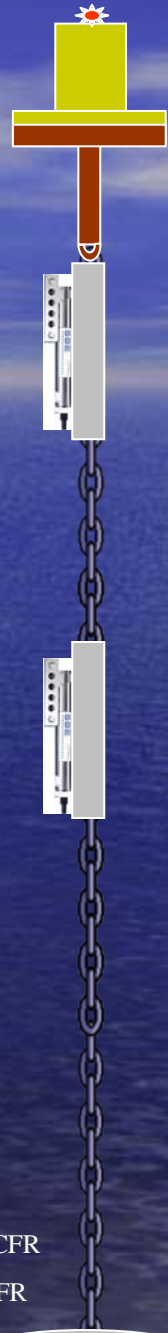


# Samantha Miller





# Mooring Design



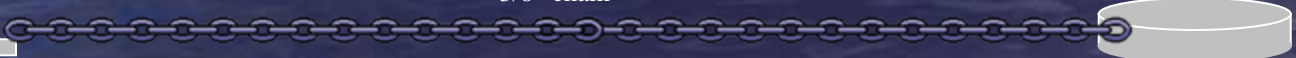
15m depth – 30 m chain

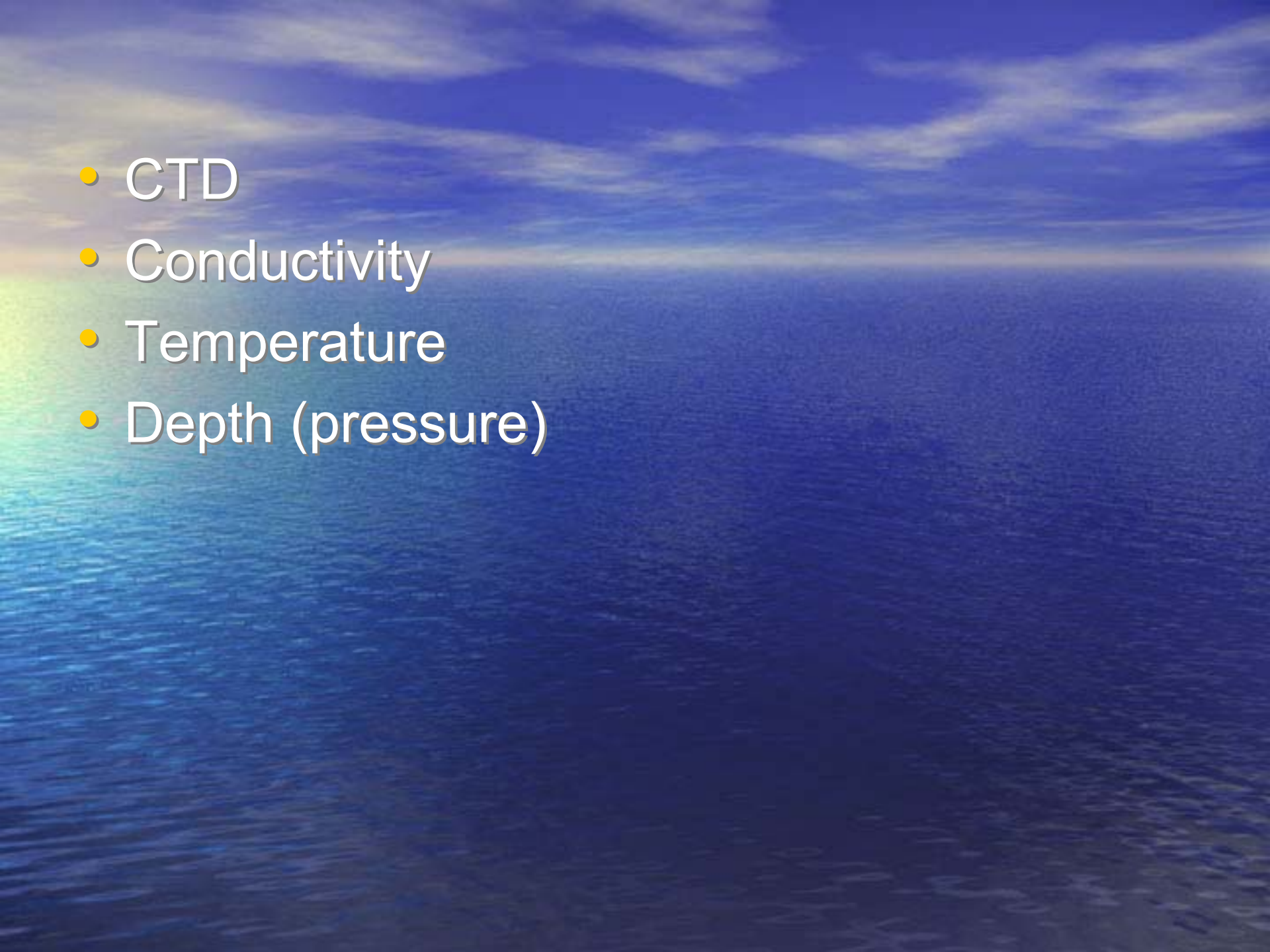
20m depth – 40 m chain

3/8" chain

1000 lb. for 4 CFR

500 lb. for 5 CFR



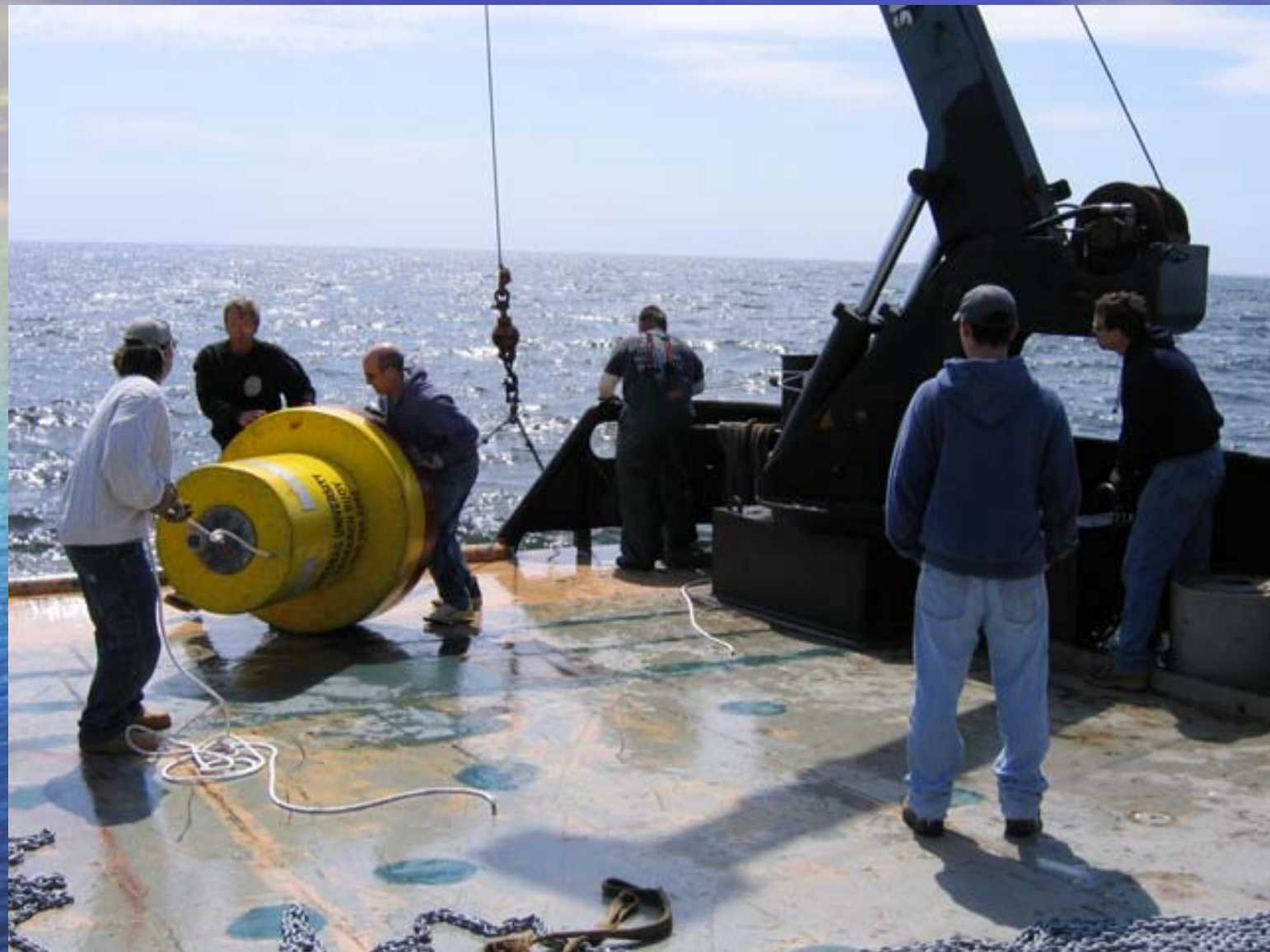
- 
- CTD
  - Conductivity
  - Temperature
  - Depth (pressure)











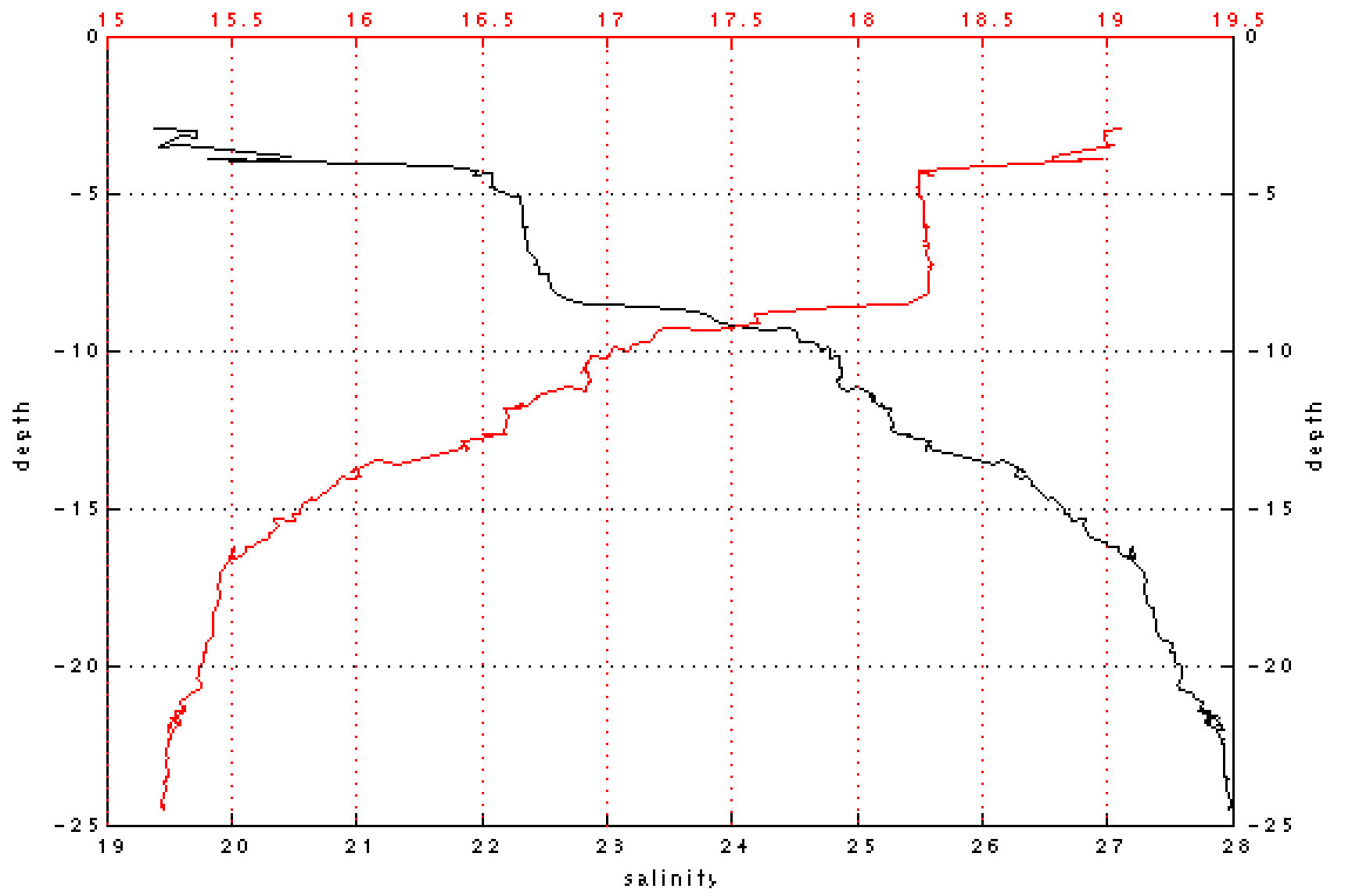


Official USCG Photo courtesy Jim Flynn



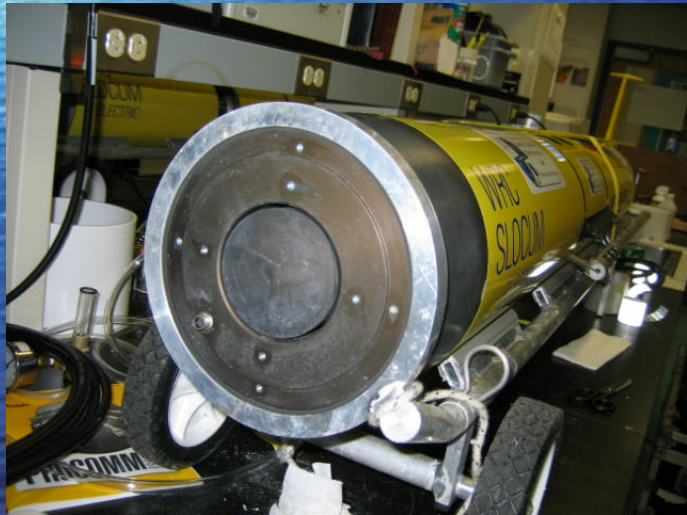
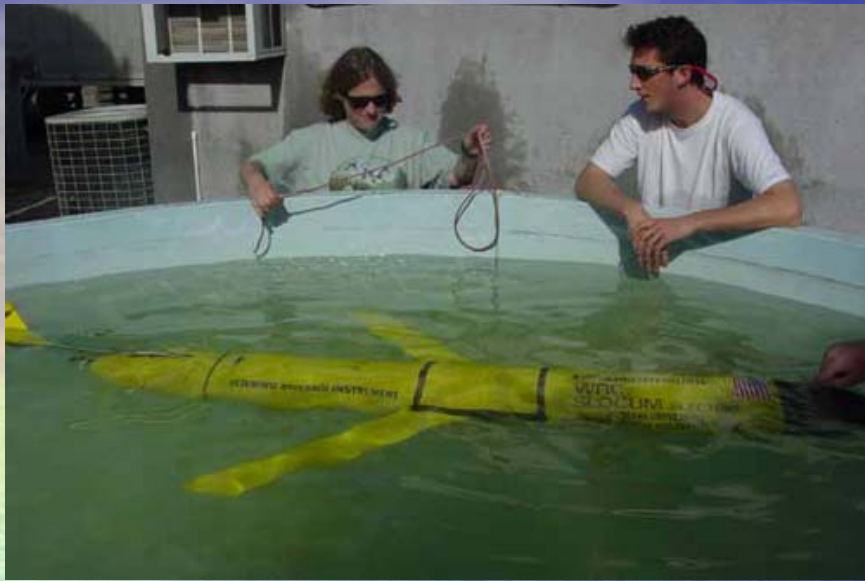
Narrows -> 40.6023 -74.0367

temperature



# Glider





- ***Glider Specifications***

- **Weight: 52 Kg** **Glider Specs**

- **Diameter: 21.3 cm**

- **Length: 1.5 meters**

- **Depth Range: 4 - 200 meters**

- **Speed: 0.4 m/sec**

- **Energy: Alkaline Batteries**

- **Endurance: ~30 days**

- **Range: 1500 km**

- **Navigation: GPS, internal dead reckoning, altimeter**

- **Sensor Package: Conductivity, Temperature, Depth**

- **Communications: satellite telephone, Freewave Radio**

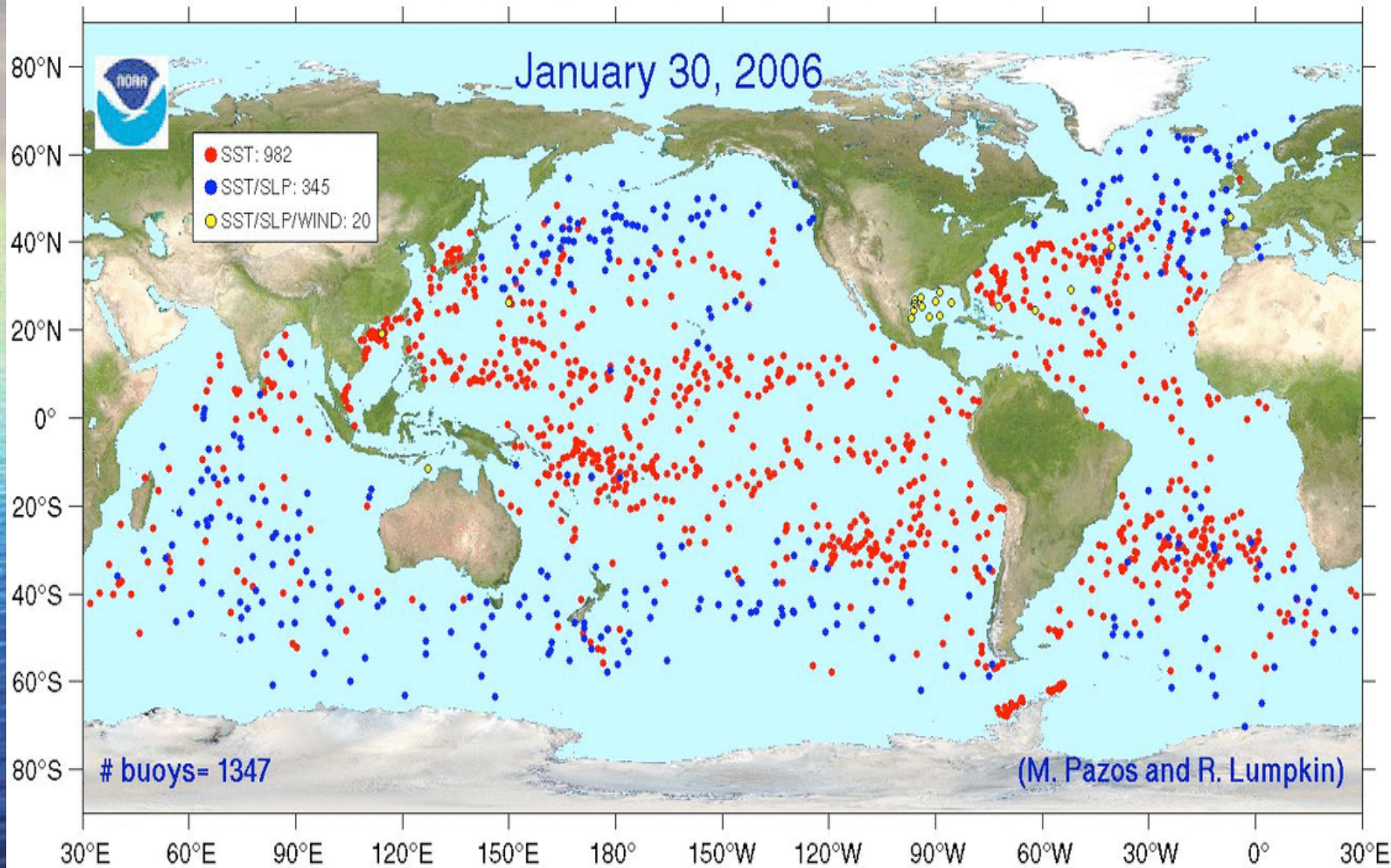
ARGOS satellite





# STATUS OF GLOBAL DRIFTER ARRAY

January 30, 2006



Aquarius





# Requirements

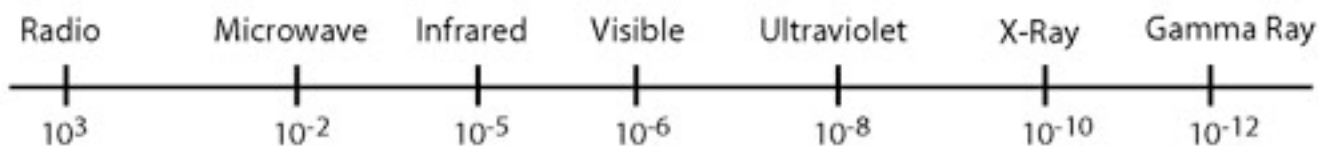
- 390 km swath
- 8 samples per month near equator
- 0.2 psu (0.0002) accuracy after temporal averaging over a month
- Temporal Resolution 1 month
- Spatial Resolution 100km

# Instrument Specifications

- Radiometer & Scatterometer
  - Footprint size - 76x94, 84x120, 96x156
  - Orbit 657 km Sun Synchronous
- Radiometer
  - Frequency  $\nu = 1.413 \text{ GHz} \pm 25\text{MHz}$
  - Wavelength  $\lambda = c/\nu \sim 0.212\text{m}$
  - L band = 0.39 - 1.55 GHz

## THE ELECTRO MAGNETIC SPECTRUM

Wavelength  
(metres)



Frequency  
(Hz)



# Scatterometer

- Frequency  $\nu = 1.26 \text{ GHz}$
- Wavelength  $\lambda = c/\nu \sim 0.238\text{m}$

# Orbit Altitude vs. Orbit Time

$$F_{\text{centrifugal}} = F_{\text{gravity}}$$

$$F_g = GMm/r^2$$

$$F_c = ma = m \Delta v / \Delta t$$

$$d = v \Delta t$$

$$t = 2\pi r / v$$

$$t = 2\pi v / a$$

$$r/v = v/a$$

$$a = v^2/r$$

$$F_c = m v^2/r = GMm/r^2$$

$$v^2 = GM/r \text{ (no sat. mass)}$$

$$v = 2\pi r / t$$

$$4\pi^2 r^2 / t^2 = GM/r$$

$$t^2 = 4\pi^2 r^3 / (GM)$$



# Orbit Altitude vs. Orbit Time

$$t^2 = 4\pi^2 r^3 / (GM)$$

$$G = 6.67E-11 \text{ Nm}^2 \text{ kg}^{-2} \quad M_e = 6E24 \text{ kg}$$

$$N = \text{kg m s}^{-2} \quad R_e = 6378 \text{ km}$$

mean altitude of satellite = 657km

$$r = 6378\text{km} + 657\text{km} = 7035\text{km} \square$$

$$t^2 = \frac{4\pi^2 (7035\text{km} (1000\text{m/km}))^3}{6.67E-11 \text{ kg m s}^{-2} \text{ m}^2 \text{ kg}^{-2} 6E24 \text{ kg}}$$

$$t \sim 97 \frac{2}{3} \text{ minutes}$$

- Ocean remote sensing from space is the only practical way to obtain global coverage of ocean surface properties
- In Situ (place) measurements important for below surface measurements and for validating satellites

# AQUARIUS / SAC-D

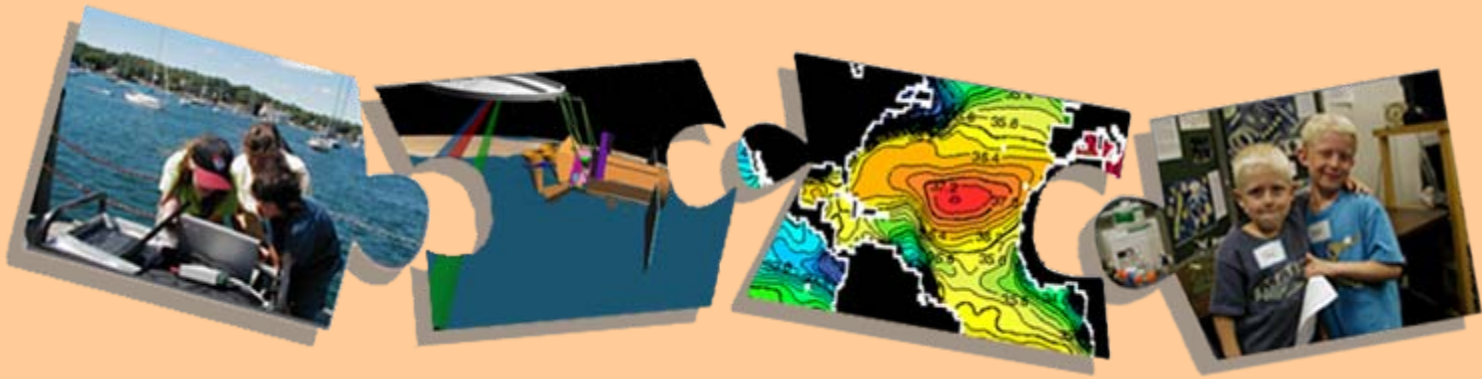
EDUCATION & PUBLIC OUTREACH

Will present two activities:

- Hands-on lab
- Online data tools

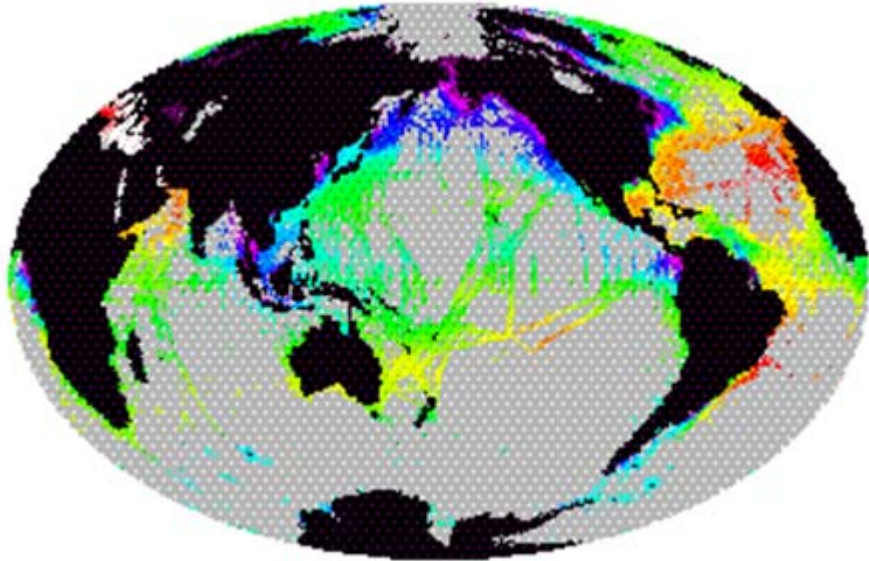
Goal is to demonstrate how these complementary activities support this "Big Idea":

*Salt causes water to freeze at a lower temperature. The oceans do not freeze -- and are layered -- because of salinity (and this is critically important to the overall earth system).*



# AQUARIUS / SAC-D

EDUCATION & PUBLIC OUTREACH

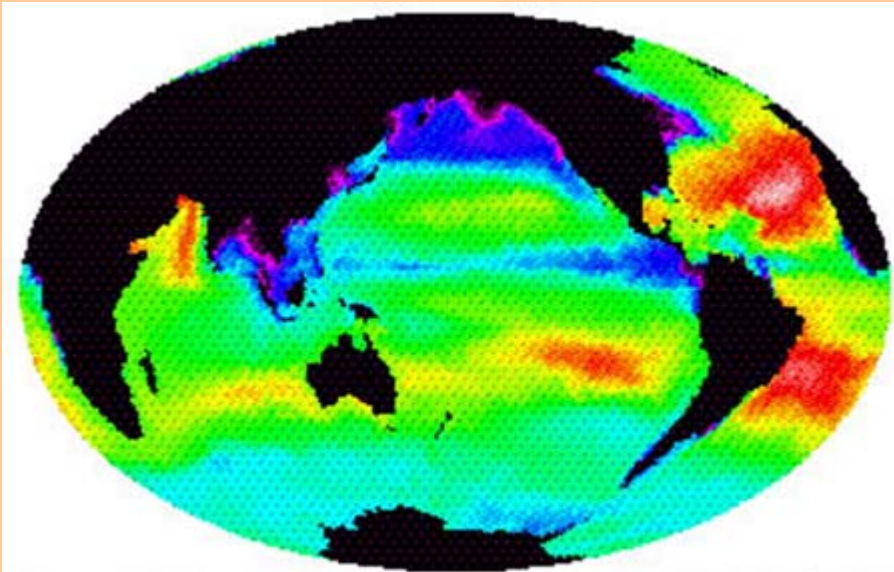


*100 years of Sea Surface Salinity (SSS) measurements*

This map shows a seven-day average, based on all known measurement locations sampled over the past 100 years for Sea Surface Salinity (SSS)

# AQUARIUS / SAC-D

EDUCATION & PUBLIC OUTREACH



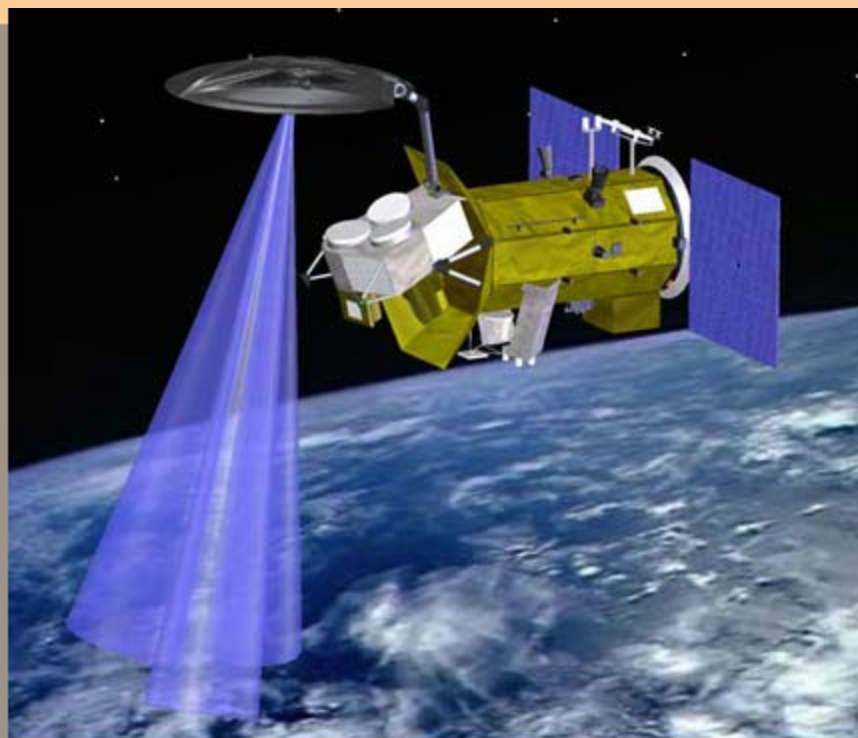
Example 7 days of Aquarius Sea Surface Salinity (SSS) data

Not presently measured by satellite, this map shows the weekly SSS coverage that will be achieved by *Aquarius* after its 2010 launch

*Aquarius / SAC-D* is a joint U.S. (NASA) - Argentine (CONAE) venture

# AQUARIUS / SAC-D

EDUCATION & PUBLIC OUTREACH



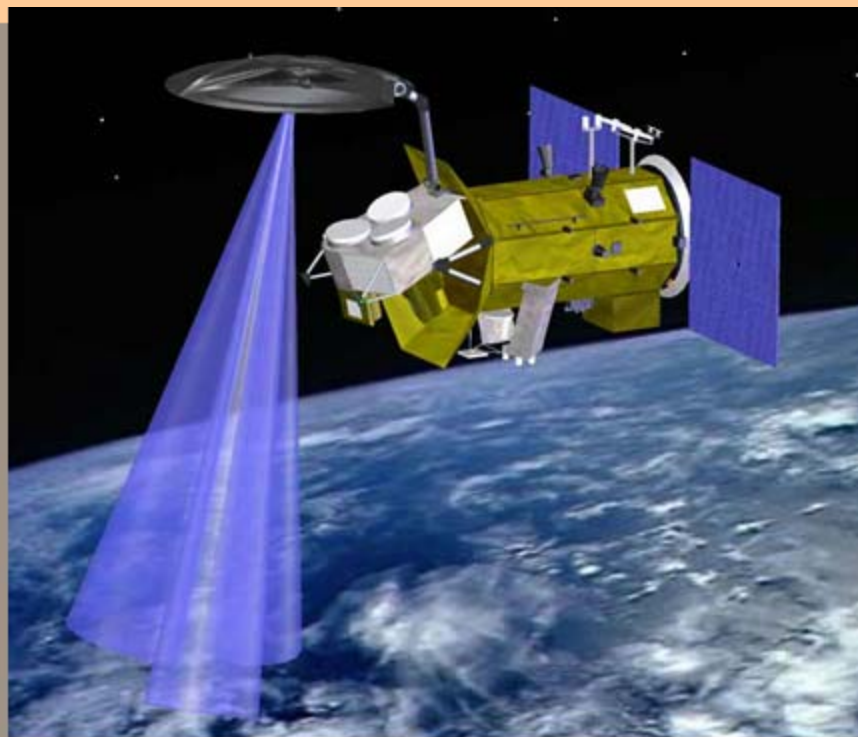
*Aquarius* instrument accuracy will be 0.2 psu\*. How much salt should be added to 1 gallon of water to change its salinity by 0.2 psu?

- 1 tablespoon
- 1 teaspoon
- 1/2 teaspoon
- 1/6 teaspoon

*\*Used to describe the concentration of dissolved salts in water, the UNESCO Practical Salinity Scale of 1978 defines salinity in terms of a conductivity ratio, so it is dimensionless. Salinity was formerly expressed in terms of parts per thousand or by weight (parts per thousand or ‰). That is, a salinity of 35 ppt meant 35 pounds of salt per 1,000 pounds of seawater. Open ocean salinities are generally in the range between 32 and 37.*

# AQUARIUS / SAC-D

EDUCATION & PUBLIC OUTREACH



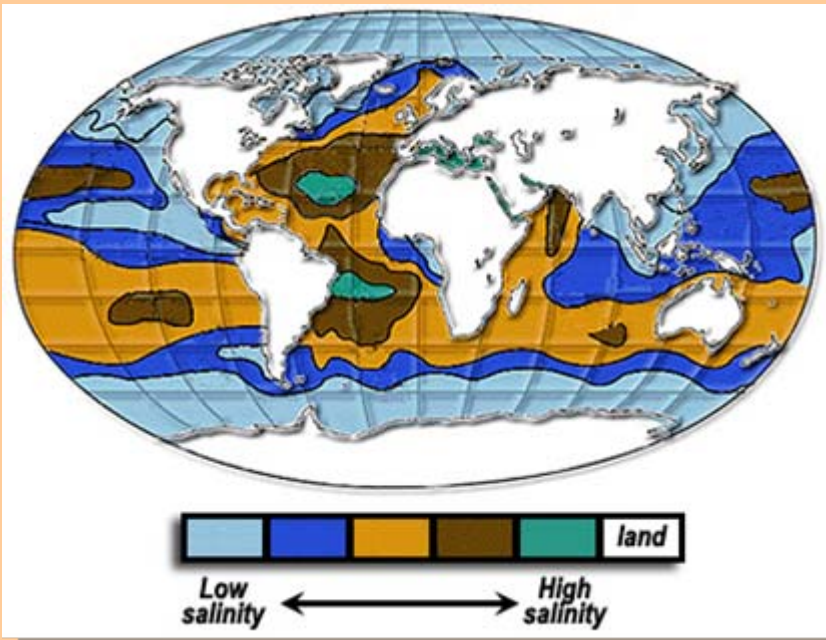
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# AQUARIUS / SAC-D

EDUCATION & PUBLIC OUTREACH



The mission will help answer questions about how our oceans respond to climate change and the water cycle.

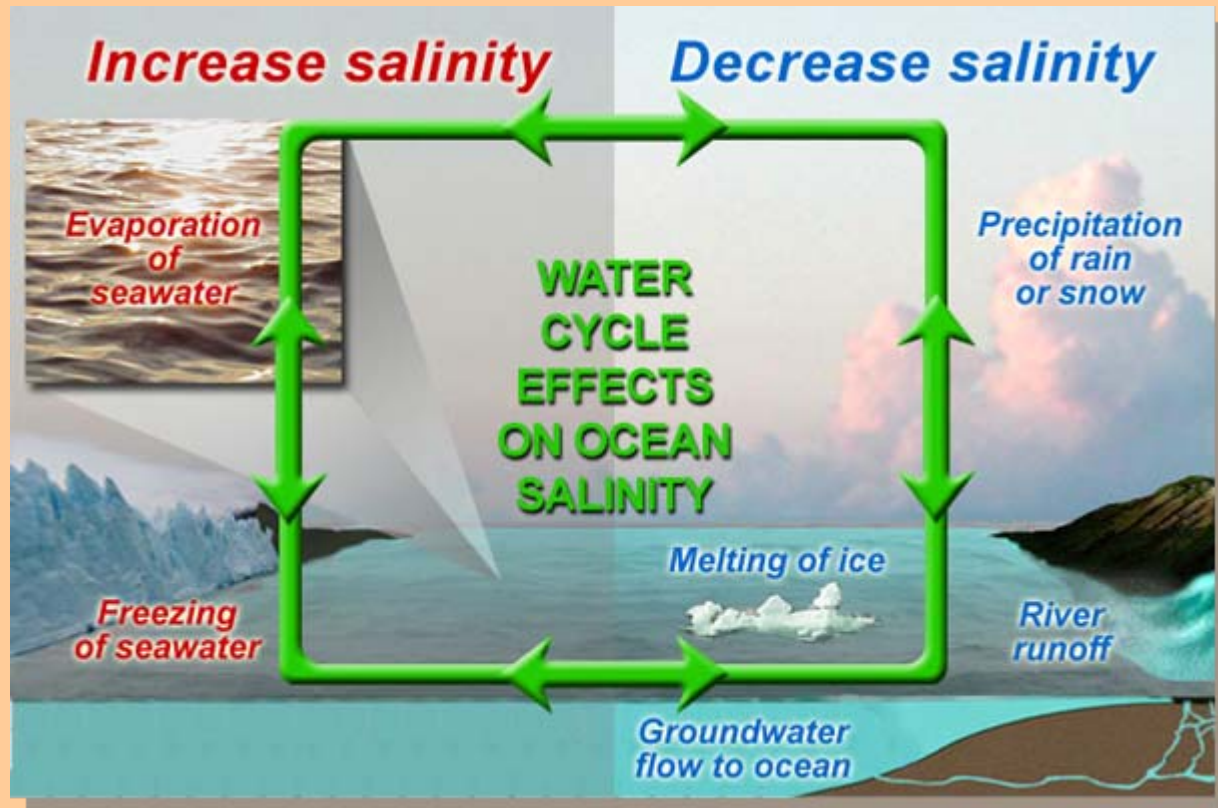
Like on land, some areas of the ocean are rainy whereas others are arid. *Aquarius* SSS data will reveal the water cycle's ever-changing "fingerprint."



# AQUARIUS / SAC-D

EDUCATION & PUBLIC OUTREACH

SSS maps can be used to directly track variations in the water cycle such as river runoff, sea ice freezing & melting, ocean evaporation & precipitation:



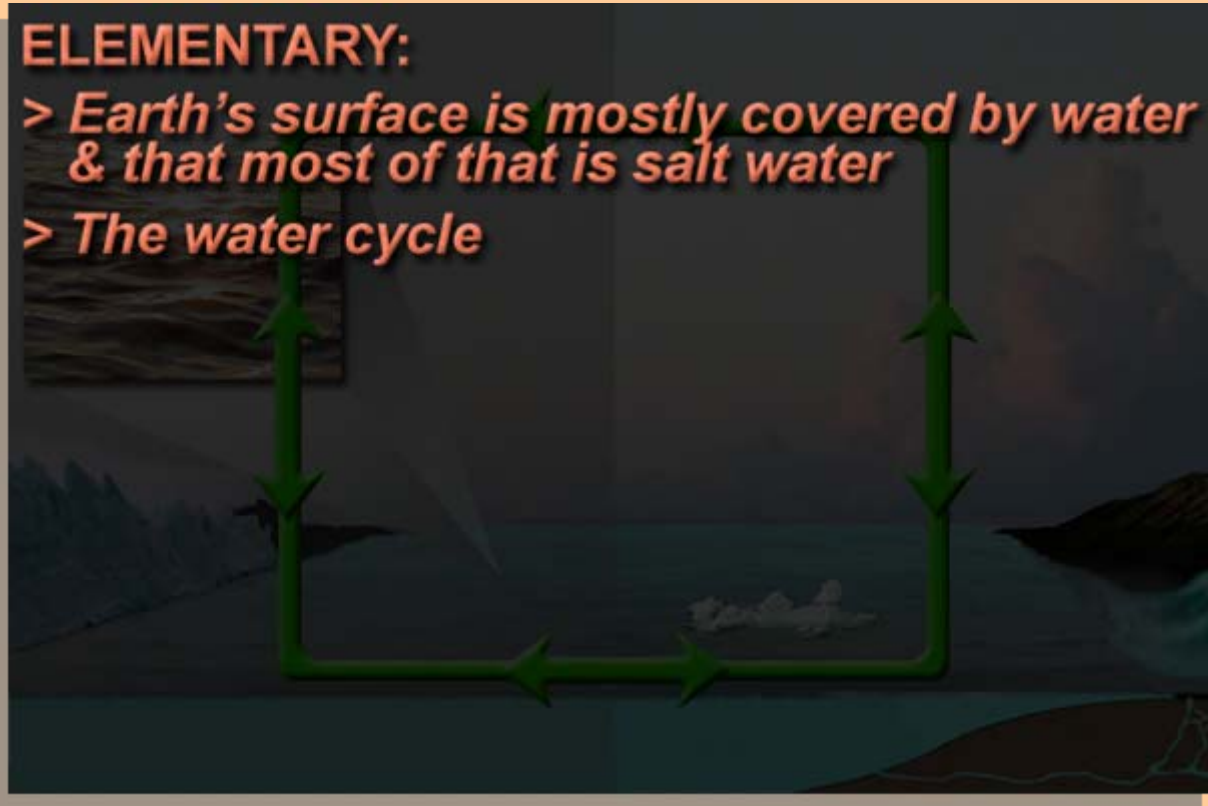
# AQUARIUS / SAC-D

EDUCATION & PUBLIC OUTREACH

These concepts are directly aligned with content knowledge Standards & Benchmarks:

## **ELEMENTARY:**

- > Earth's surface is mostly covered by water & that most of that is salt water**
- > The water cycle**



# AQUARIUS / SAC-D

EDUCATION & PUBLIC OUTREACH

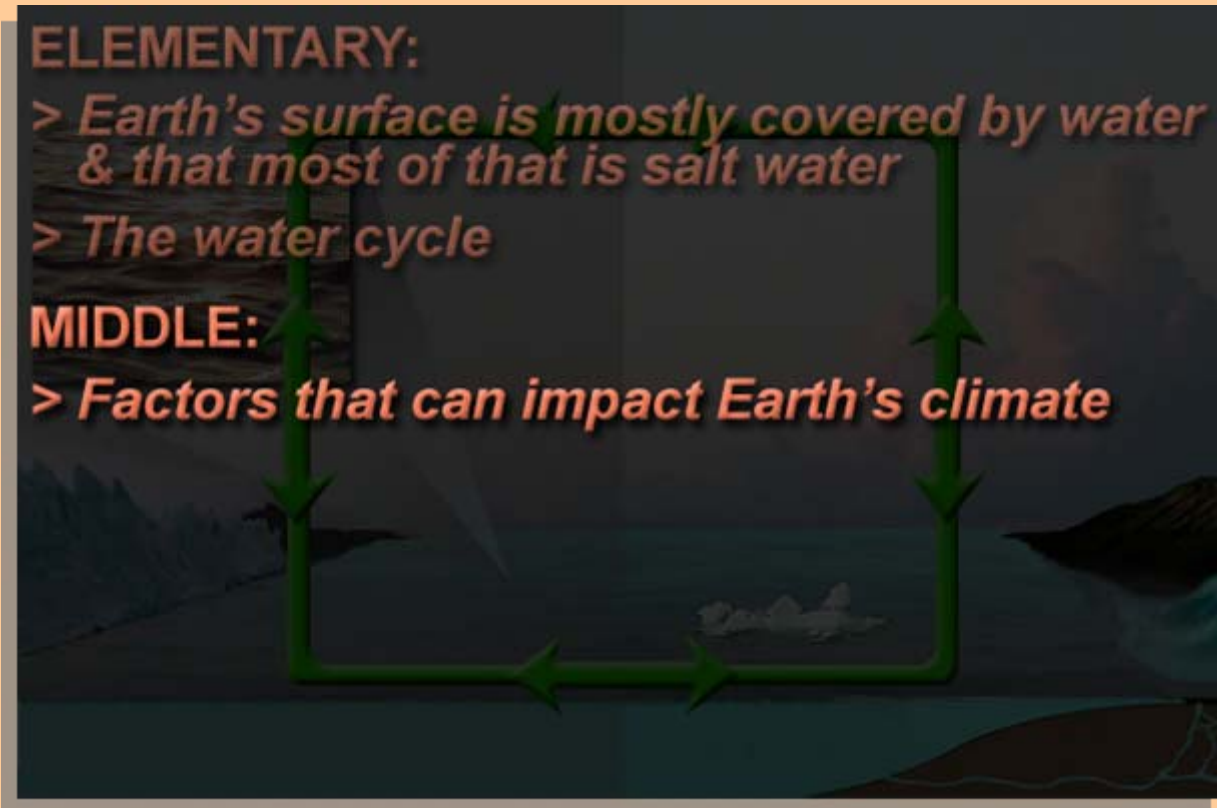
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- > *The water cycle*

## MIDDLE:

- > *Factors that can impact Earth's climate*



# AQUARIUS / SAC-D

EDUCATION & PUBLIC OUTREACH

These concepts are directly aligned with content knowledge Standards & Benchmarks:

## ELEMENTARY:

- > *Earth's surface is mostly covered by water & that most of that is salt water*
- > *The water cycle*

## MIDDLE:

- > *Factors that can impact Earth's climate*

## HIGH:

- > *The effects of different physical cycles on the environment of Earth*
- > *How physical processes affect different regions of the world*

# AQUARIUS / SAC-D

## EDUCATION & PUBLIC OUTREACH

“Salinity Patterns & the Water Cycle” module recently rated “Very Good-Outstanding” by NASA Earth Science Education Products Review panel.

Includes nine activities aligned with physical sciences standards for elementary through high school students.

<http://aquarius.nasa.gov/education.html>

**AQUARIUS**  
Sea Surface Salinity from Space

HOME | OVERVIEW | SCIENCE | TECHNOLOGY & OPERATIONS | GALLERY | EDUCATION | NEWSROOM | LINKS | SITE MAP

EDUCATION: CLASSROOM ACTIVITIES

+ Salinity Basics + Classroom Activities + Salinity Data & Tools + Links

### CAN SEAWATER FREEZE?

Grade Level: Middle | Time: Four 45 min class periods | Content: NSES Physical Science, Ocean Literacy Principle 1a

**Big Idea**  
Salt causes water to freeze at a lower temperature. In fact, any substance dissolved in water lowers its freezing point. The oceans do not freeze (except in extreme polar areas) because of the salinity (i.e., concentration of salts).

**Key Concepts**

- Dissolving any substance in pure water raises or lowers the freezing and boiling point.
- When water freezes - goes from the liquid state to the solid state - its molecules go from a disorganized state to an organized state.
- When water freezes to a solid, molecular motion slows down enough that the molecules become permanently fixed.

Time	Starting salt / water content	FREEZER EXPERIMENT					
		1 hour in freezer		24 hours in freezer		48 hours in freezer	
		Temp	Notes	Temp	Notes	Temp	Notes
A							
B							
C							
D							

**BUOYANCY EXPERIMENT**

	Ice from "A"	Ice from "B"	Ice from "C"	Ice from "D"
Buoyant "A"				
Buoyant "B"				
Buoyant "C"				
Buoyant "D"				

**Sea ice photos from the Arctic**

**Iceberg photos from the Antarctic**

Photos from Aquarius Experiments for Student Activities and the Aquarius Education Journal

# AQUARIUS / SAC-D

## EDUCATION & PUBLIC OUTREACH

Each activity follows the format:

- Big Idea
- Key Concepts
- Essential Questions
- Knowledge and Skills
- Prior Knowledge
- Common Preconceptions

And is linked to the comprehensive concept map: “Water & its patterns on Earth’s surface”

**AQUARIUS**  
Sea Surface Salinity from Space

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• Salinity Basics • Classroom Activities • Salinity Data & Tools • Links

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- Dissolving any substance in pure water raises or lowers the freezing and boiling point.
- When water freezes - goes from the liquid state to the solid state - its molecules go from a disorganized state to an organized state.
- When water freezes to a solid, molecular motion slows down enough that the molecules become permanently fixed in an orderly arrangement called a crystal.
- The individual particles that make up salt (known as ions) arrange themselves around the water molecules. In doing so, they shield the water molecules from interactions among themselves, making it less likely that they will find each other and form ice.
- The water molecules have to be slowed down even more in the presence of salt in order to form a solid. Do you

ID	Starting salt / water content	Record your observations below:					
		1 hour in freezer		24 hours in freezer		48 hours in freezer	
A		Temp	Notes	Temp	Notes	Temp	Notes
B							
C							
D							

	Ice from "A"	Ice from "B"	Ice from "C"	Ice from "D"
Solution "A"				
Solution "B"				
Solution "C"				
Solution "D"				

Sea ice photos from the Arctic

Iceberg photos from the Antarctic

# AQUARIUS / SAC-D

## EDUCATION & PUBLIC OUTREACH

"Can Sea Water Freeze?" is a middle school activity

- Let's begin the experiment!

FREEZER EXPERIMENT							
Jar	Starting salt / water content	Record your observations below:					
		1 hour in freezer		24 hours in freezer		48 hours in freezer	
		Temp.	Notes	Temp.	Notes	Temp.	Notes
A							
B							
C							
D							

BUOYANCY EXPERIMENT				
	Ice from "A"	Ice from "B"	Ice from "C"	Ice from "D"
Solution "A"				
Solution "B"				
Solution "C"				
Solution "D"				

Sea ice photos from the Arctic

Iceberg photos from the Antarctic

## AQUARIUS

Education & Public Outreach

[HOME](#) | [OVERVIEW](#) | [SCIENCE](#) | [TECHNOLOGY & OPERATIONS](#) | [GALLERY](#) | [EDUCATION](#) | [NEWSROOM](#) | [LINKS](#) | [SITE MAP](#)

### EDUCATION: Classroom Activities

## SEA WATER FREEZE

Grade Level: Middle | Time: 3 class periods | Content: Physical Science

#### Principles & Student Learning Objectives

- Substances have characteristic properties (e.g., density) that are independent of sample amount
- Observe how salinity affects the time it takes water to freeze
- Understand that salt is left behind as salt water freezes, this process forms brine
- Through experimentation, learn that ice is essentially salt-free whether formed from fresh or salt water
- Learn about sea ice vs. icebergs; see that they float higher in salt water than in fresh water

**BACKGROUND:** Sea ice is frozen seawater that floats on the ocean surface. Blanketing millions of square kilometers near the North and South Poles, sea ice can form and melt during different seasons, affecting both human activity and biological habitat. Almost all Antarctic sea ice (near the South Pole) melts away and re-forms each year. On the other hand, in the Arctic (near the North Pole) some sea ice persists year after year. Icebergs are also important in polar regions. Unlike sea ice that is formed from salty seawater, icebergs are composed of ice originating from land-based glaciers that flow into the sea. Glaciers are formed by snow that accumulates on mountains over many years. Ask the students whether snow is made of saltwater or freshwater. How do they know this? (Taste.)

**Materials:** Per Student Group – one cup of ordinary table salt, tablespoon, plastic ice tray with divided watertight sections, tap water, thermometer, 8 jars (at least 8 oz.), beakers (at least 300 ml) or cut 2L plastic soda bottles, ruler, graph paper.

**Preparation:** Students will need access to a freezer. An earth globe will also be shown to the students.

#### Activity

Begin with a class discussion of the locations of the North Pole (Arctic) and South Pole (Antarctic). Ask the students which Pole is covered by land (Antarctic) and which is covered by seawater (Arctic). Ask students which Pole is more likely to have glaciers nearby (North Pole). Ask the class to hypothesize about the affect of water salinity – or amount of salt – on the formation and buoyancy of sea ice. Do they think ice formed from salt water will freeze more quickly or more slowly than fresh water? Do they think that sea ice (i.e., formed from salty water) will be more buoyant or less buoyant than ice formed from fresh water (e.g., icebergs)? Do they think that the size of sea ice or icebergs affects their buoyancy (e.g., percentage above and below the water line)?

Label 8 jars as follows: 2 marked "A"; 2 marked "B"; two marked "C"; and two marked "D." In each jar mix salt and water solutions as follows:  
 In each jar marked "A": mix 9 T salt with 1 cup water  
 In each jar marked "B": mix 6 T salt with 1 cup water

**EDUCATION**

**SALINITY BASICS**

**CLASSROOM ACTIVITIES**  
Salinity Patterns & the Water Cycle

**SALINITY DATA & TOOLS**

**LINKS**

# AQUARIUS / SAC-D

EDUCATION & PUBLIC OUTREACH

*Students in action!*





# AQUARIUS / SAC-D

EDUCATION & PUBLIC OUTREACH

*Students in action!*





AQUARIUS/SAC-D  
EDUCATION & PUBLIC OUTREACH

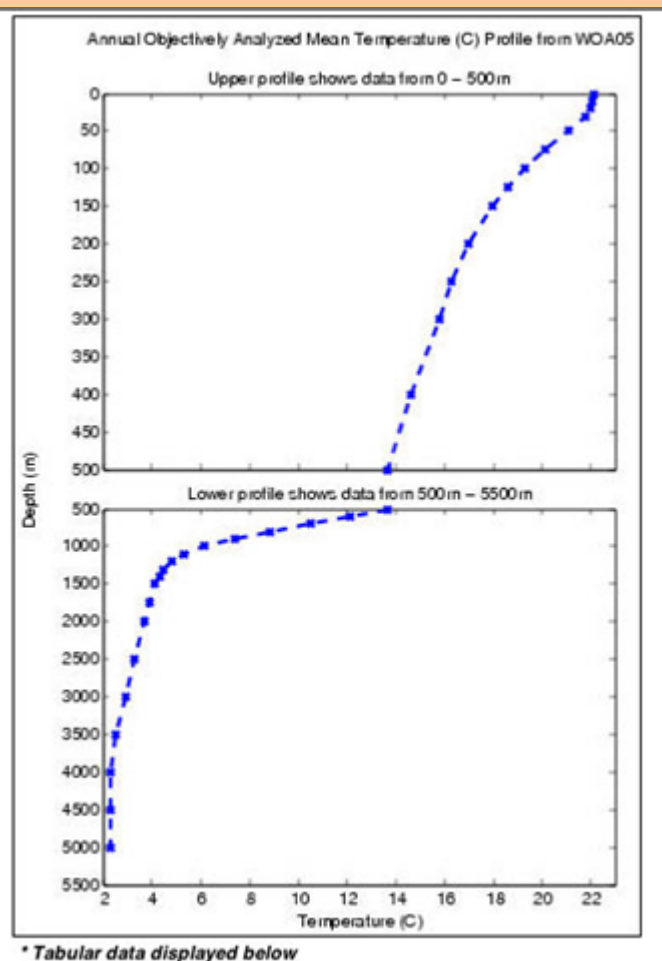
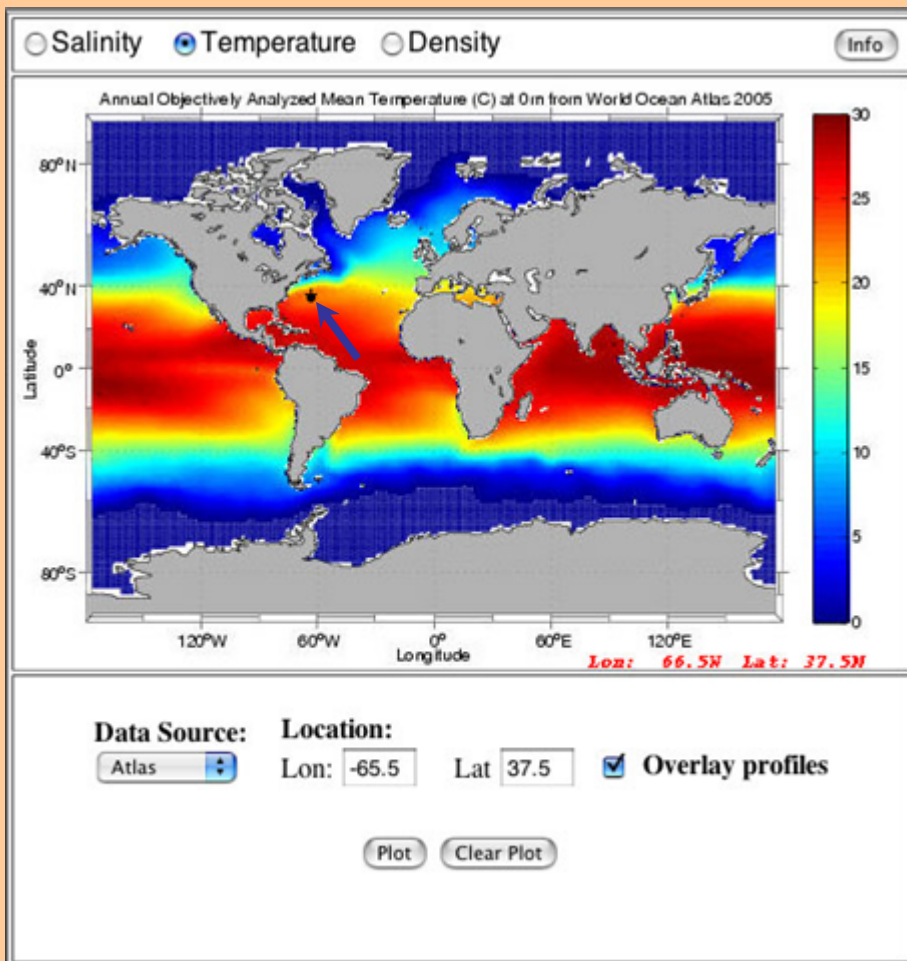
*For complementary online activities,  
let's visit the prototype website:*

*<http://aquarius.jpl.nasa.gov/>*

# AQUARIUS / SAC-D

## EDUCATION & PUBLIC OUTREACH

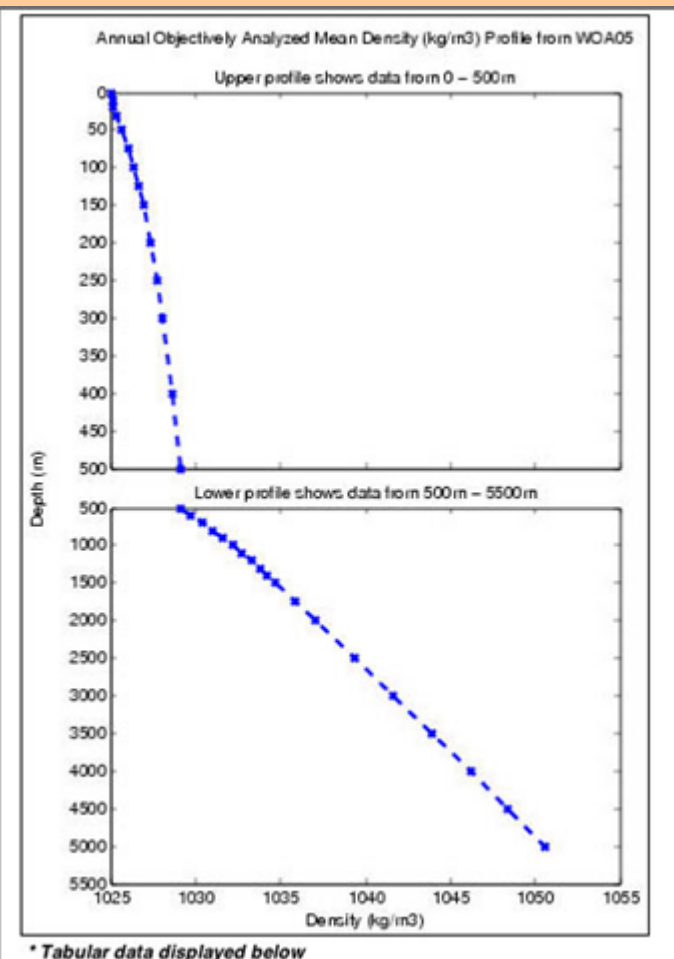
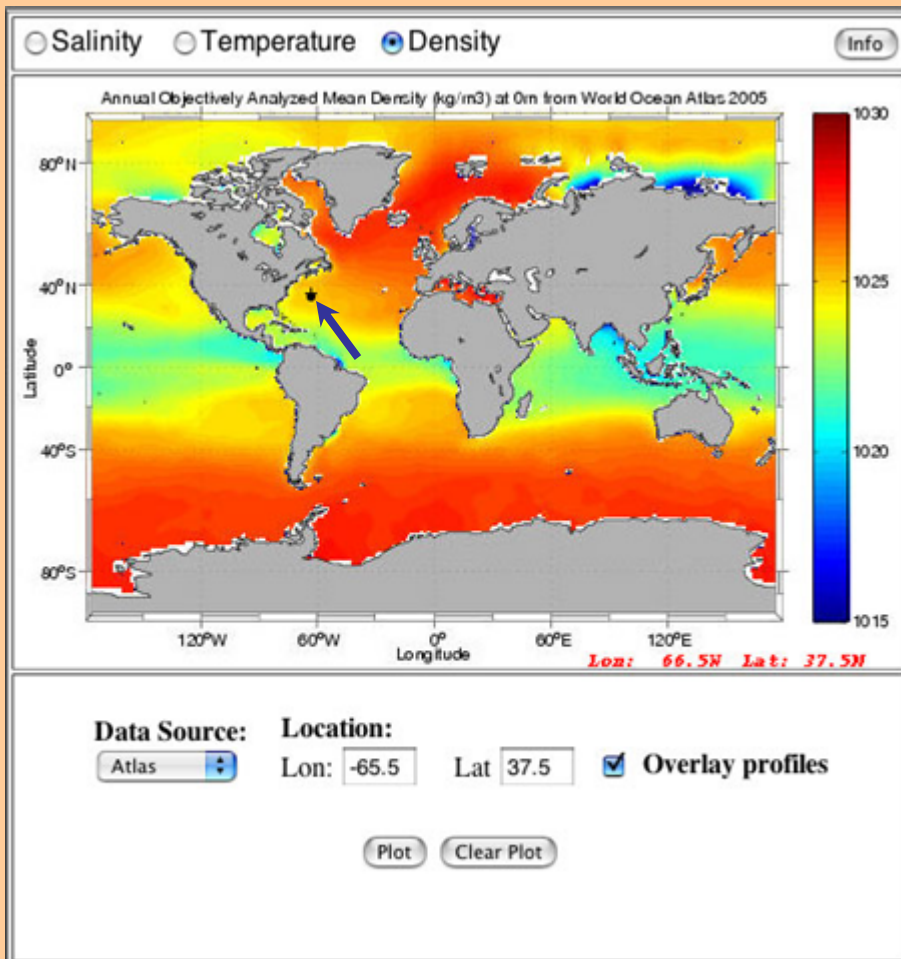
Annual mean maps & In-water profiles: Salinity, temperature & density



# AQUARIUS / SAC-D

## EDUCATION & PUBLIC OUTREACH

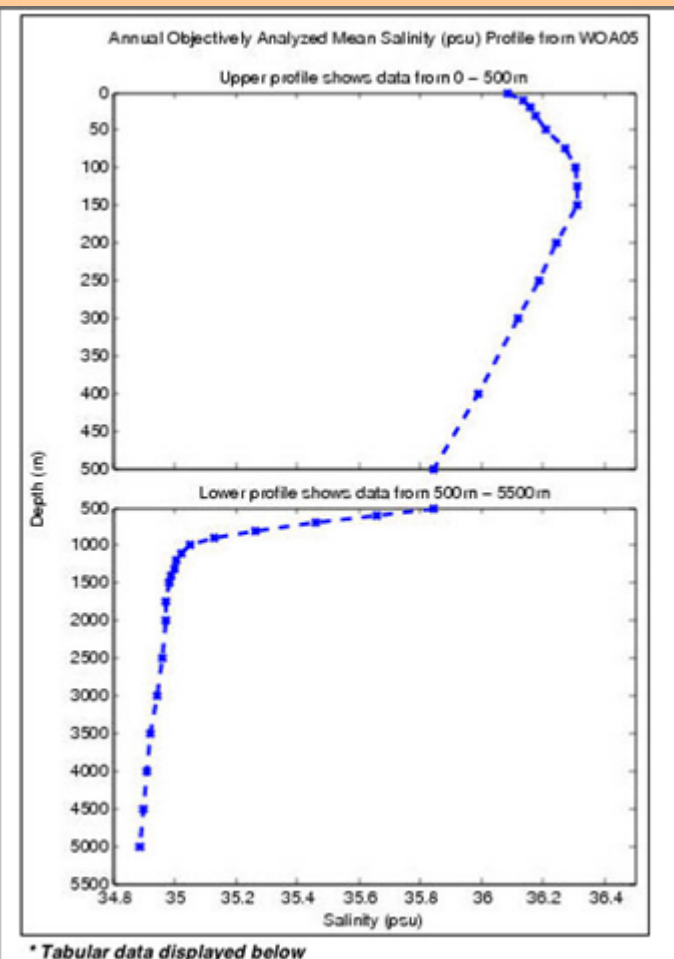
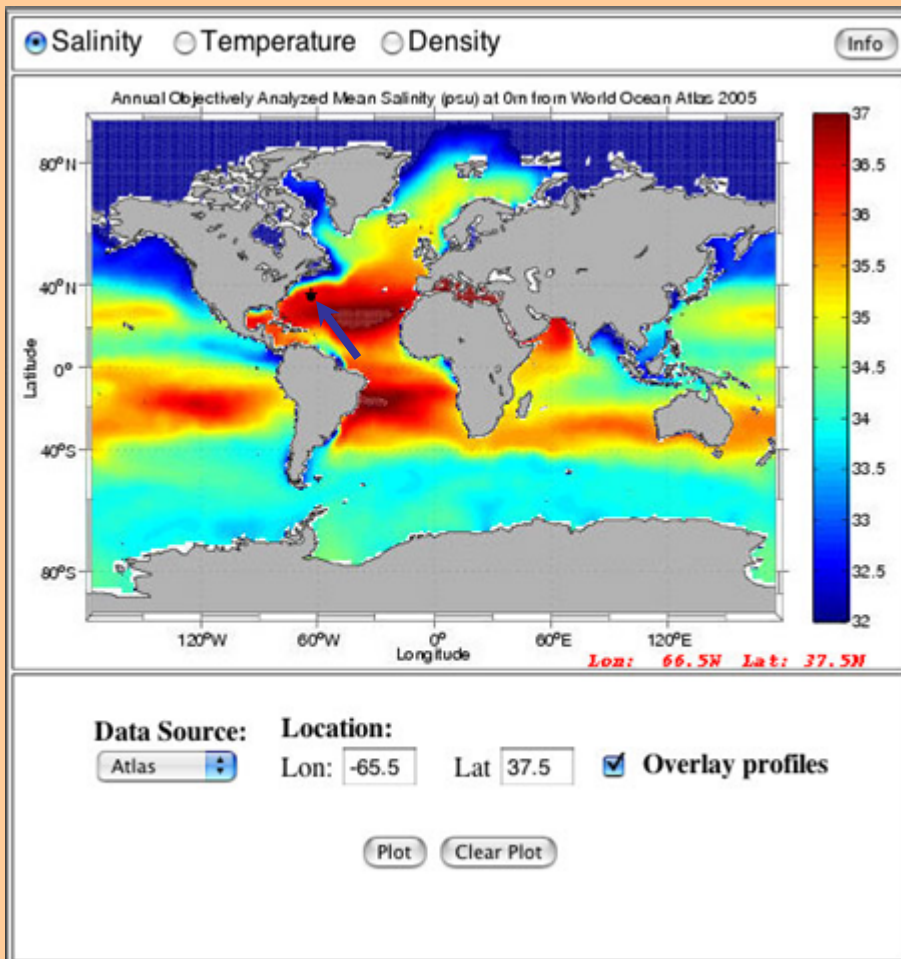
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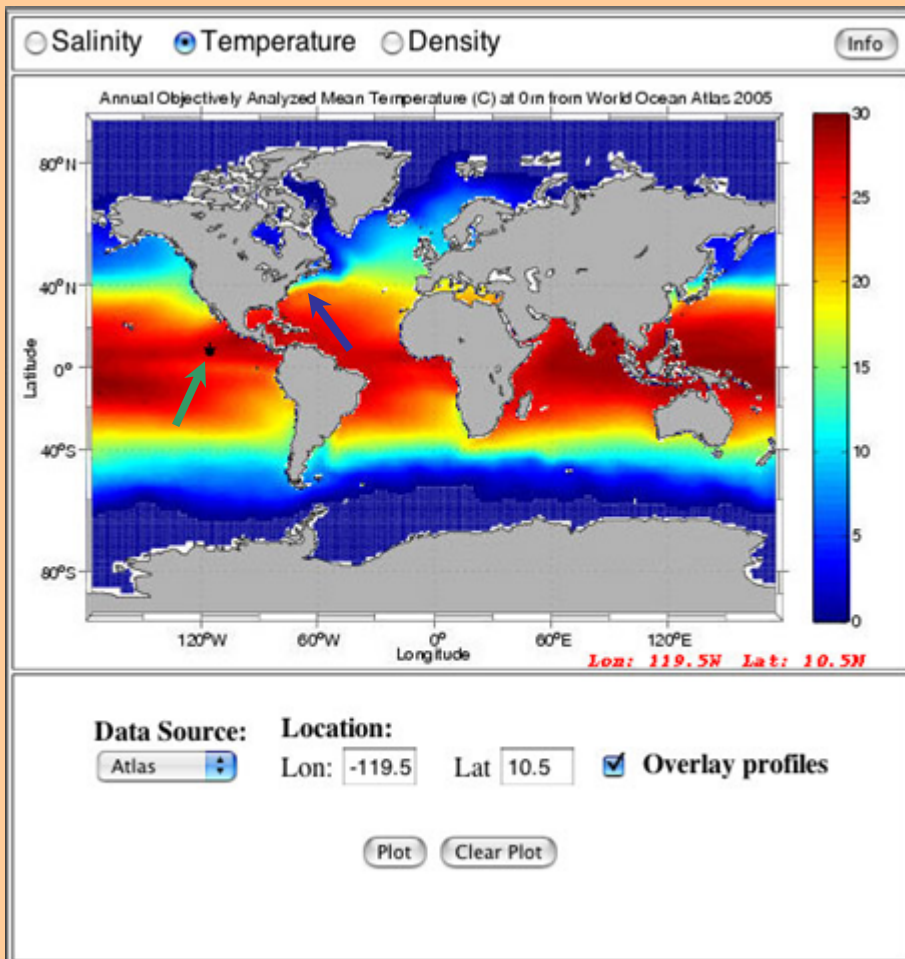
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EDUCATION & PUBLIC OUTREACH

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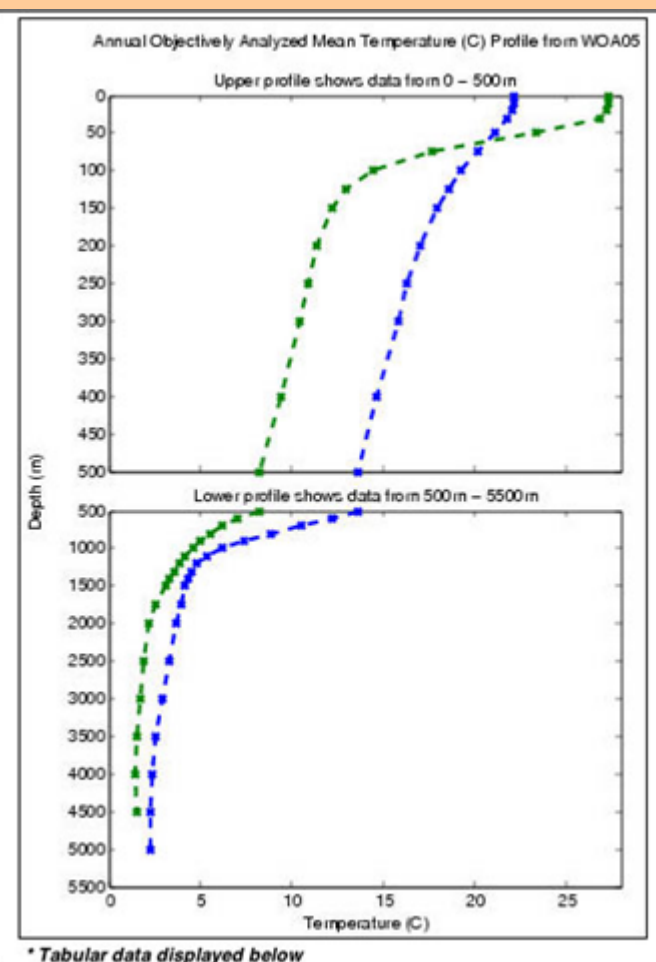
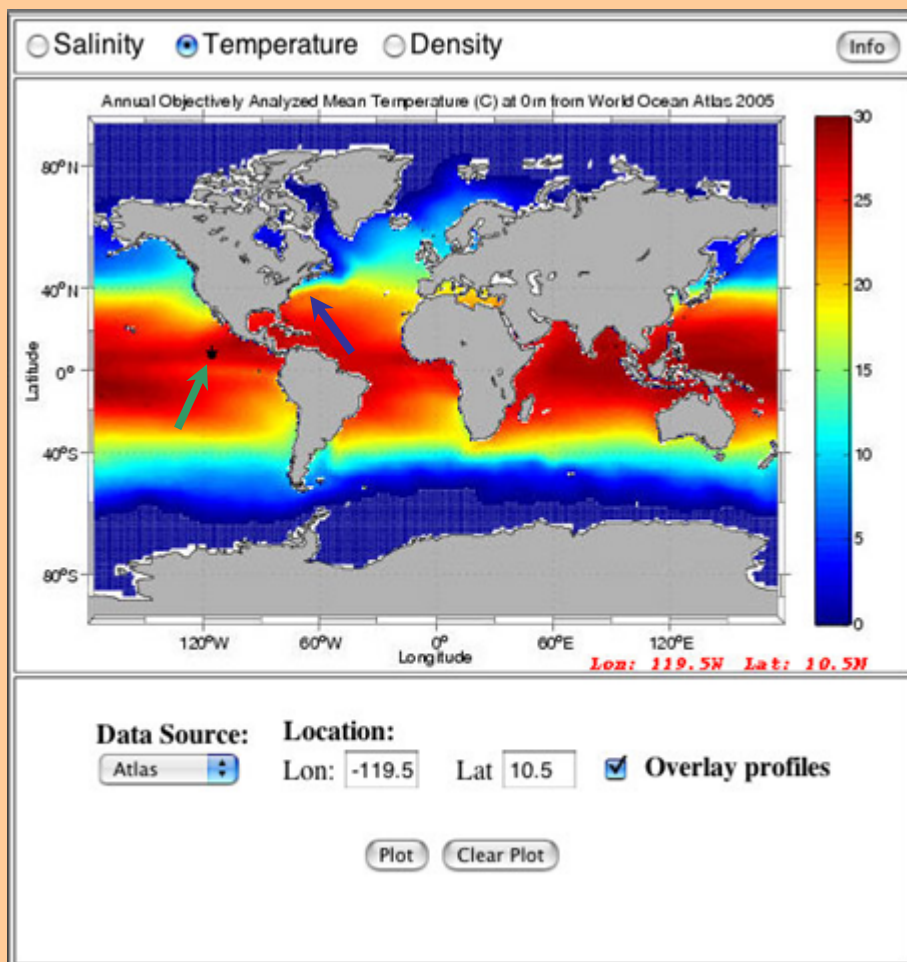


Can choose up to six locations at once

# AQUARIUS/SAC-D

## EDUCATION & PUBLIC OUTREACH

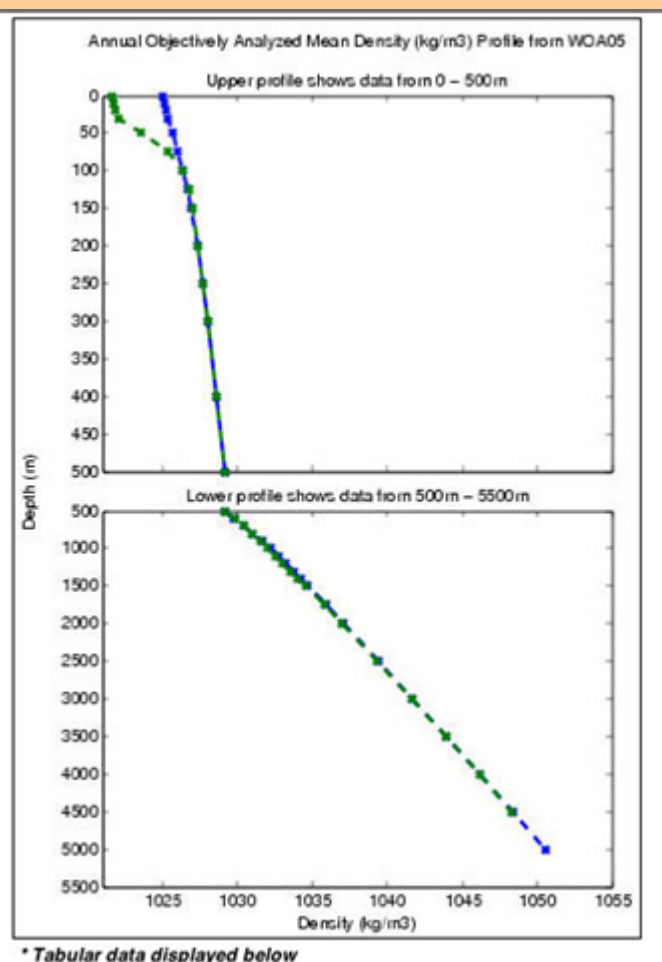
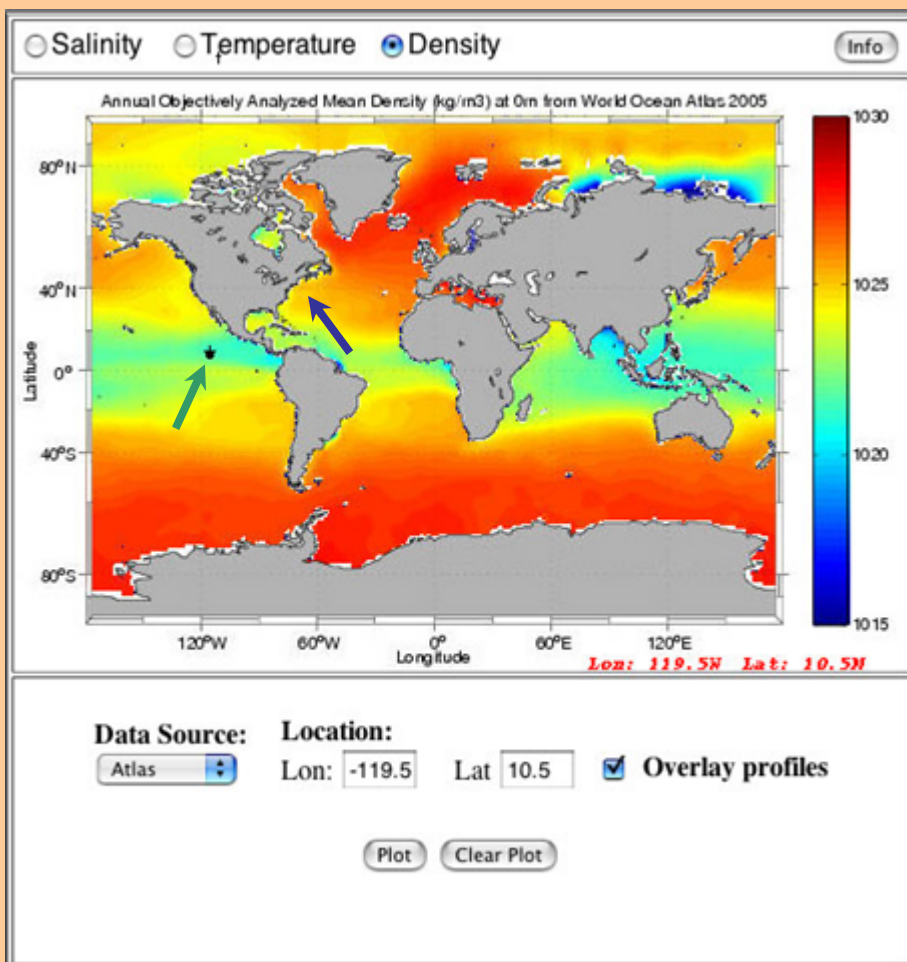
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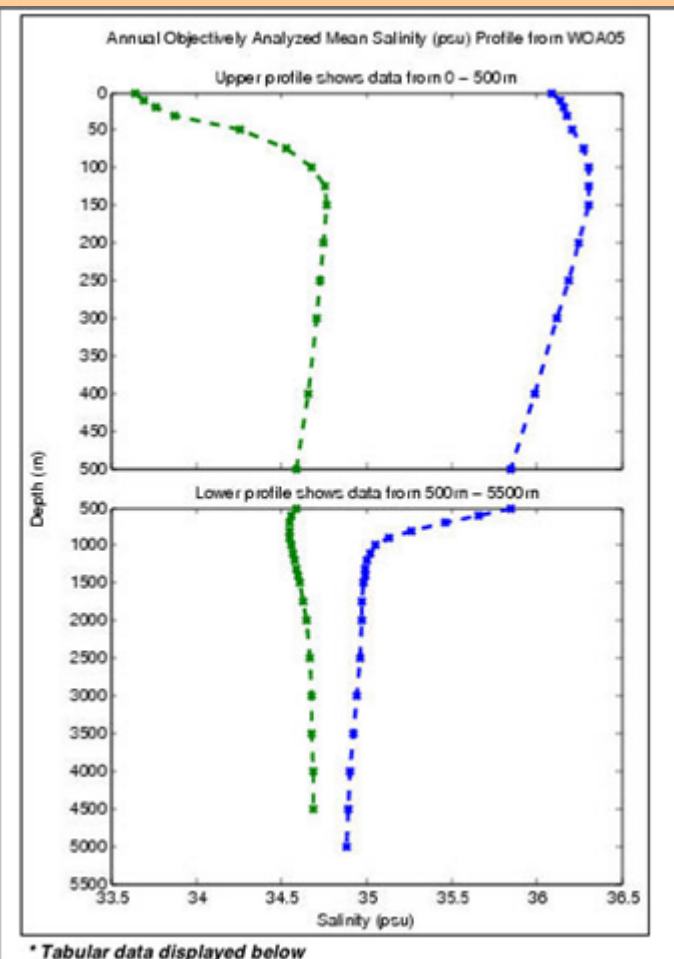
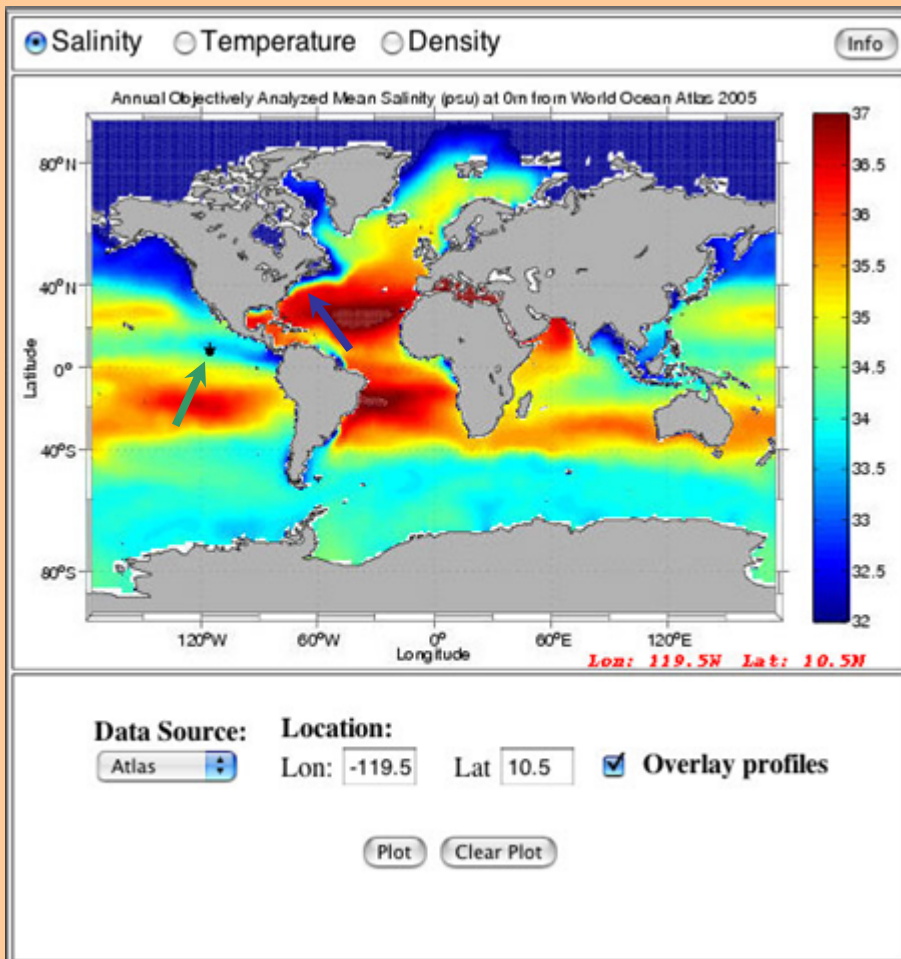




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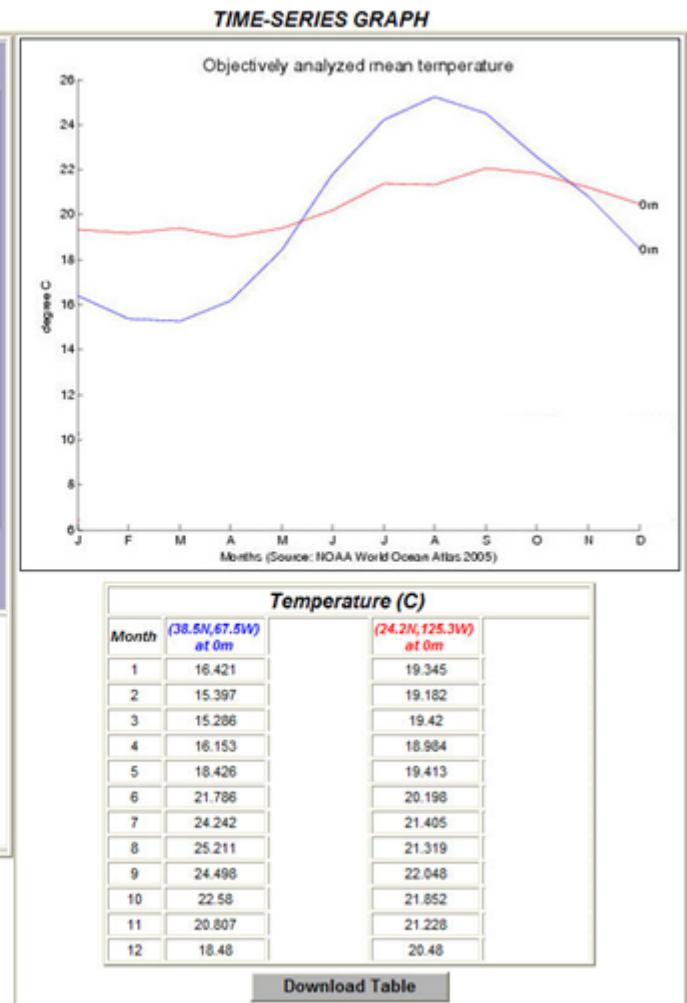
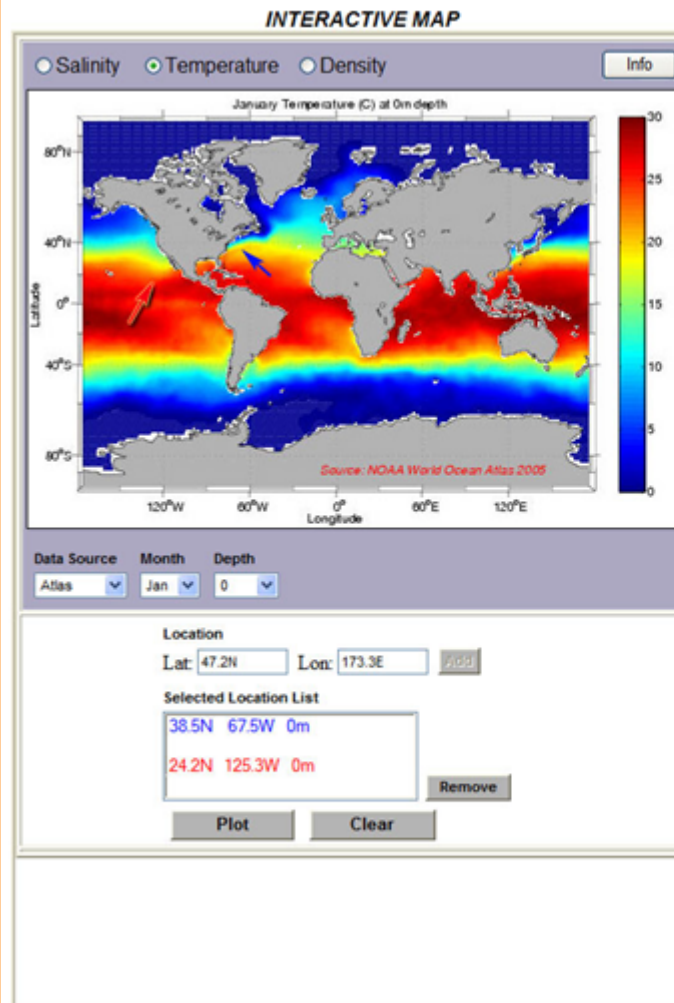


# AQUARIUS / SAC-D

## EDUCATION & PUBLIC OUTREACH

Monthly mean maps  
&  
Time-series graphs

Salinity,  
Temperature &  
Density  
from 1 - 6  
depths!

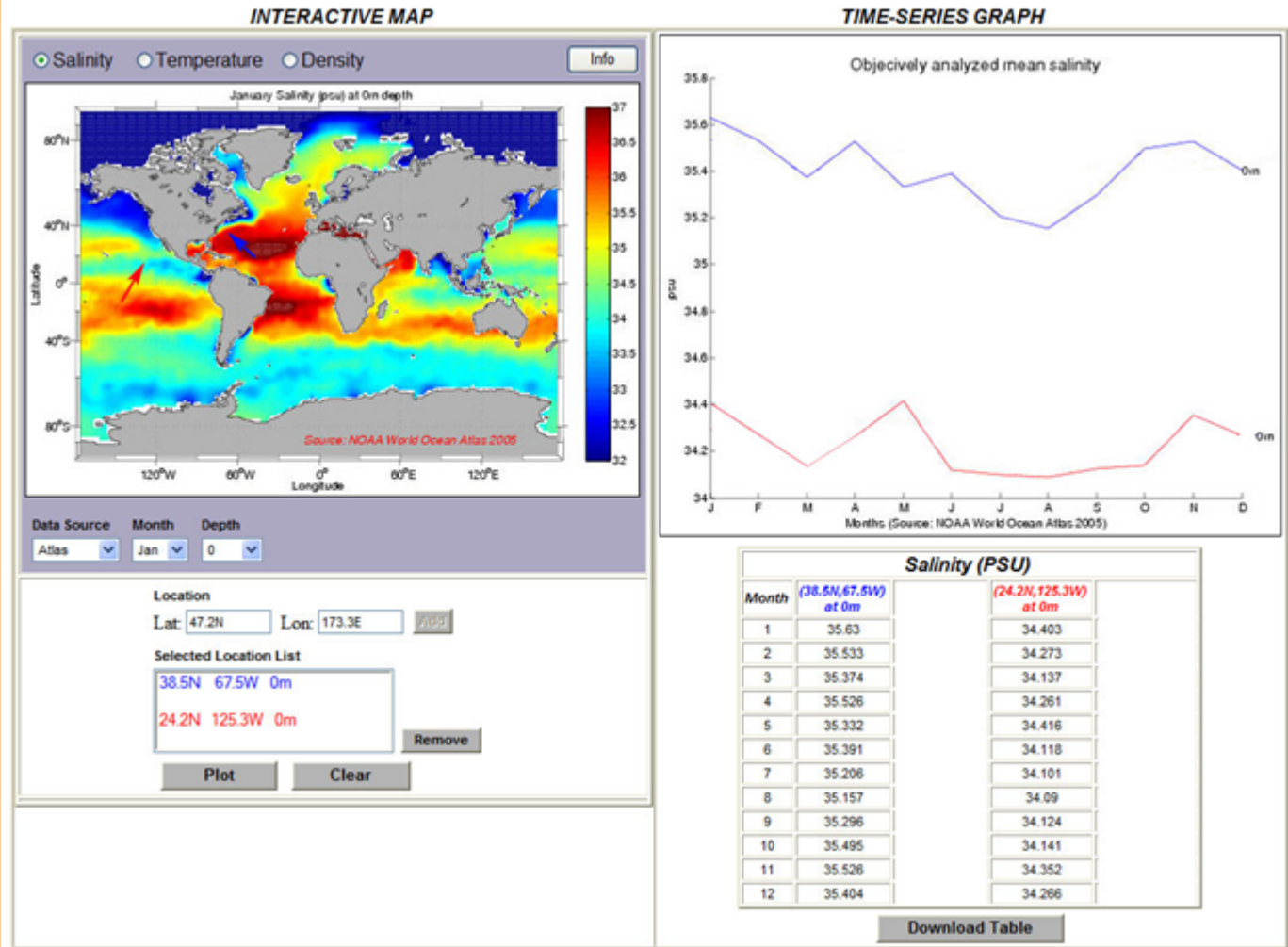


# AQUARIUS / SAC-D

## EDUCATION & PUBLIC OUTREACH

Monthly mean maps & Time-series graphs

Salinity, Temperature & Density from 1 - 6 depths!

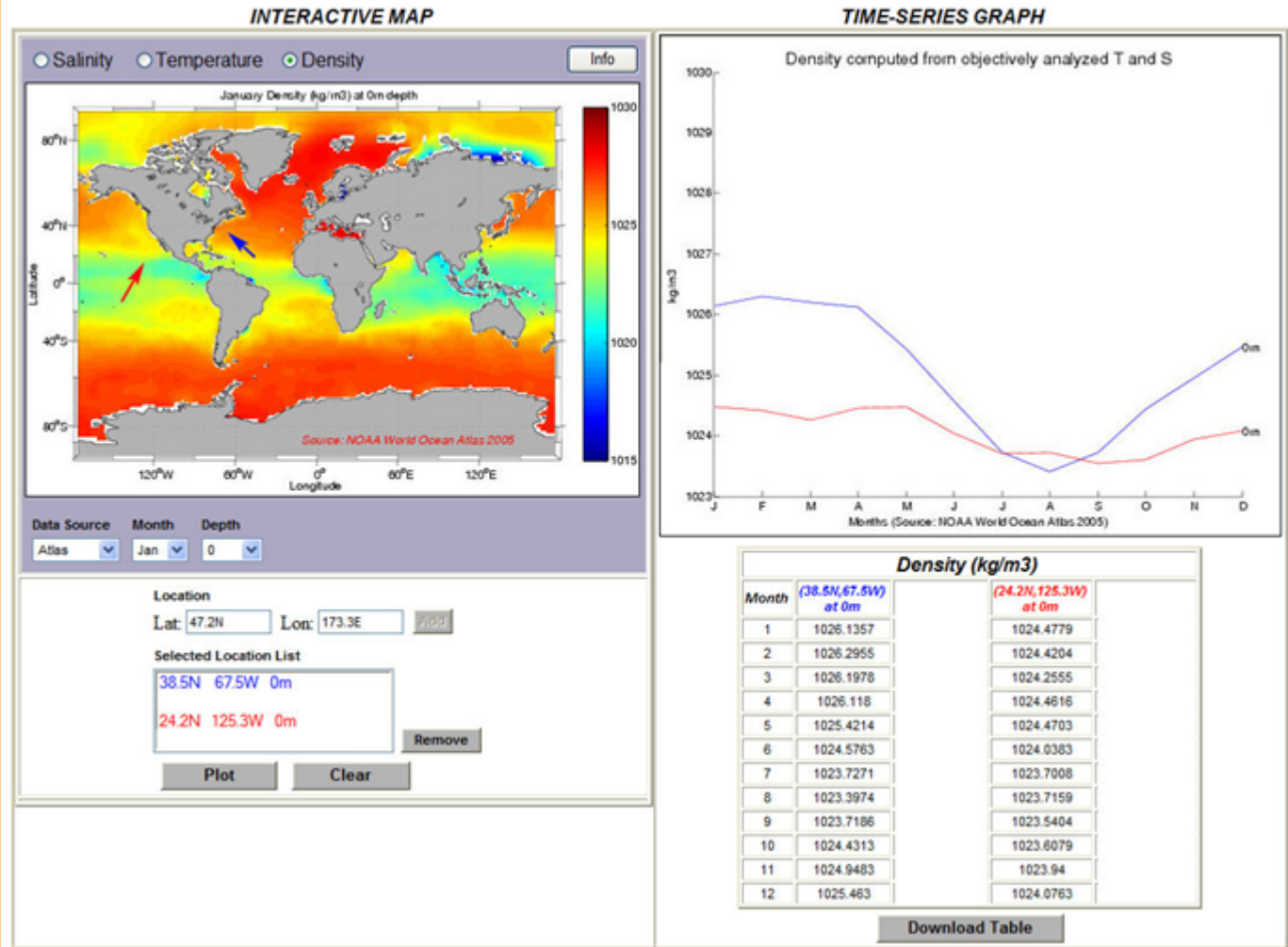


# AQUARIUS / SAC-D

## EDUCATION & PUBLIC OUTREACH

Monthly mean maps  
&  
Time-series graphs

Salinity,  
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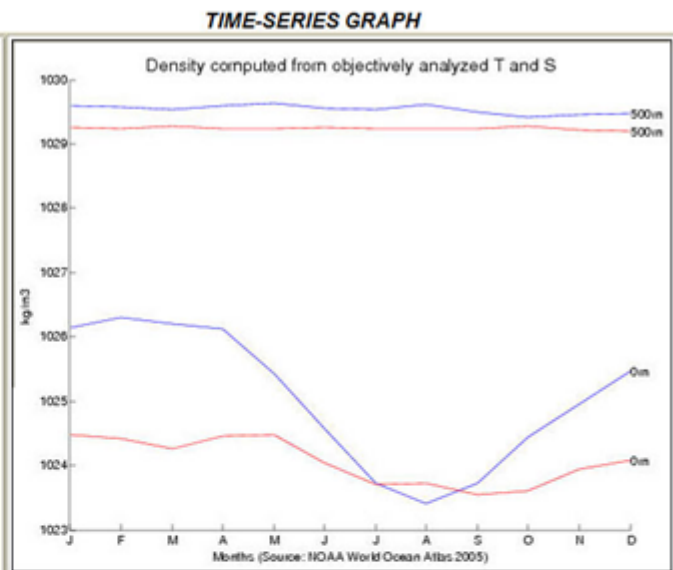
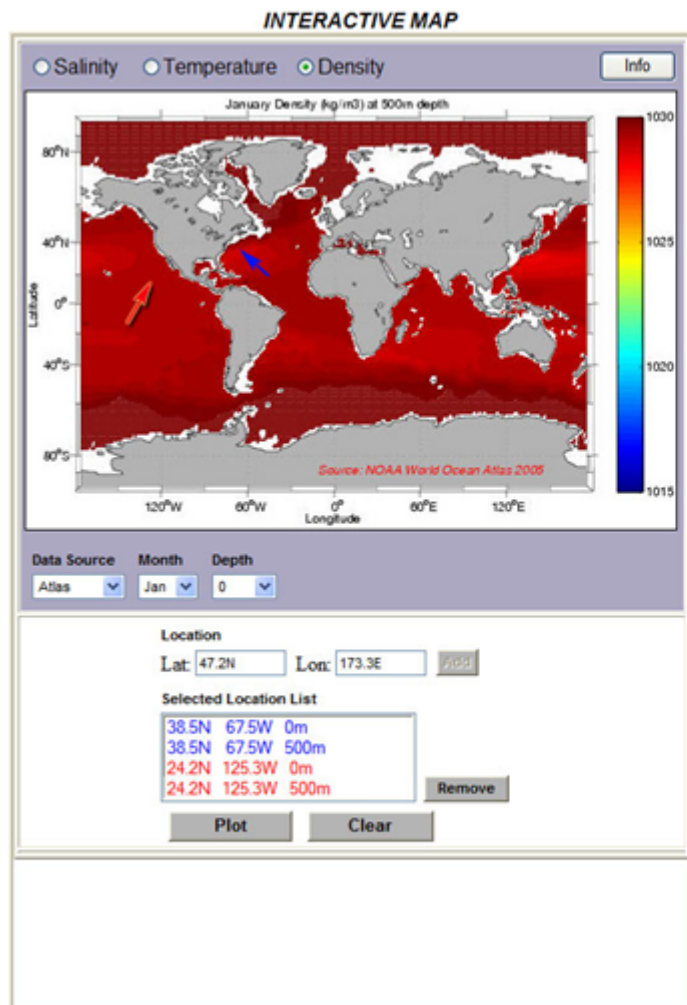


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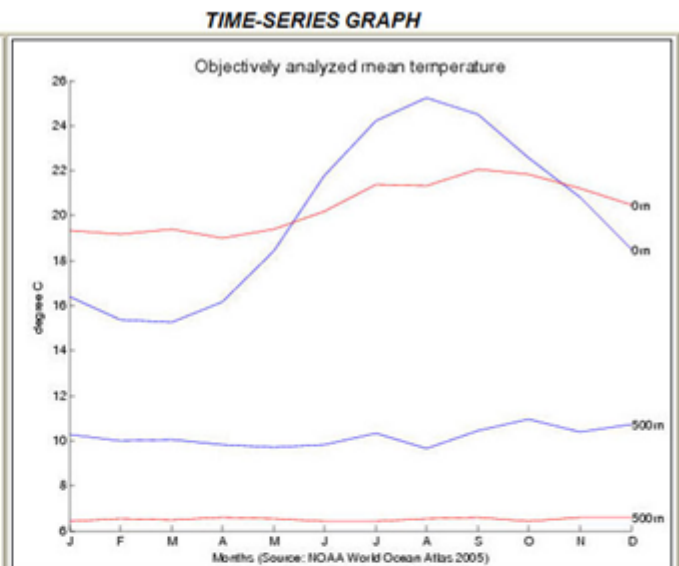
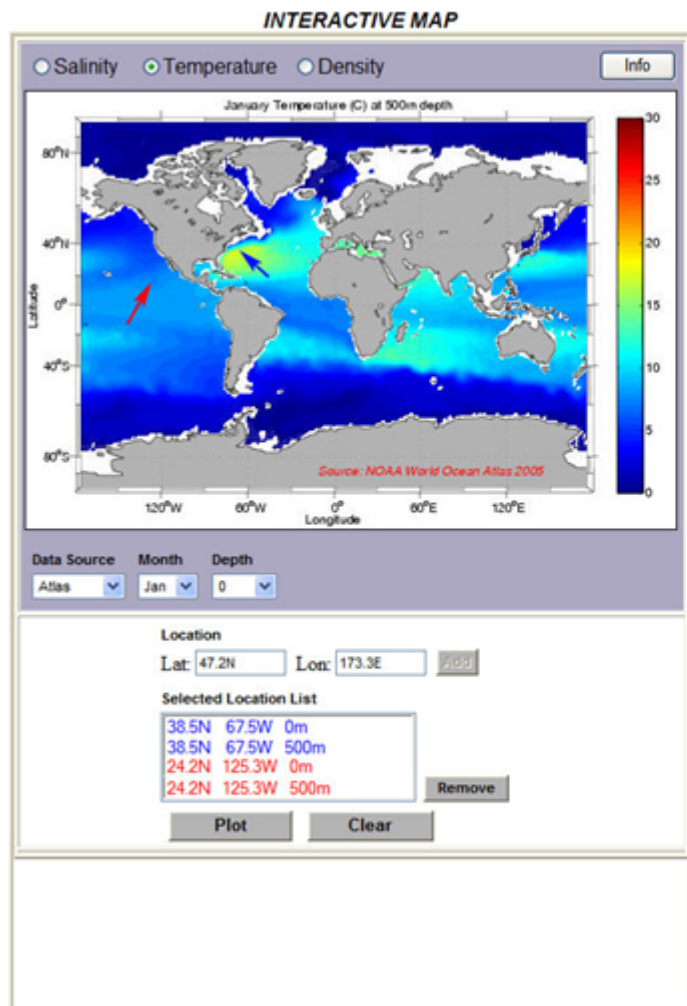
Month	Density (kg/m <sup>3</sup> )			
	(38.5N, 67.5W) at 0m	(38.5N, 67.5W) at 500m	(24.2N, 125.3W) at 0m	(24.2N, 125.3W) at 500m
1	1026.1357	1029.5865	1024.4779	1029.2524
2	1026.2955	1029.582	1024.4204	1029.2329
3	1026.1978	1029.5246	1024.2555	1029.2755
4	1026.118	1029.593	1024.4616	1029.2333
5	1025.4214	1029.636	1024.4703	1029.2438
6	1024.5763	1029.5609	1024.0383	1029.2581
7	1023.7271	1029.5249	1023.7008	1029.2375
8	1023.3974	1029.6133	1023.7159	1029.2306
9	1023.7186	1029.4933	1023.5404	1029.2448
10	1024.4313	1029.4187	1023.6079	1029.2668
11	1024.9483	1029.4641	1023.94	1029.2233
12	1025.463	1029.473	1024.0763	1029.2033

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Temperature (C)				
Month	(38.5N, 67.5W) at 0m	(38.5N, 67.5W) at 500m	(24.2N, 125.3W) at 0m	(24.2N, 125.3W) at 500m
1	16.421	10.29	19.345	6.403
2	15.397	10.005	19.182	6.545
3	15.286	10.06	19.42	6.456
4	16.153	9.837	18.984	6.575
5	18.426	9.706	19.413	6.529
6	21.786	9.838	20.198	6.446
7	24.242	10.312	21.405	6.436
8	25.211	9.672	21.319	6.517
9	24.498	10.42	22.048	6.609
10	22.58	10.931	21.852	6.434
11	20.807	10.418	21.228	6.617
12	18.48	10.723	20.48	6.592

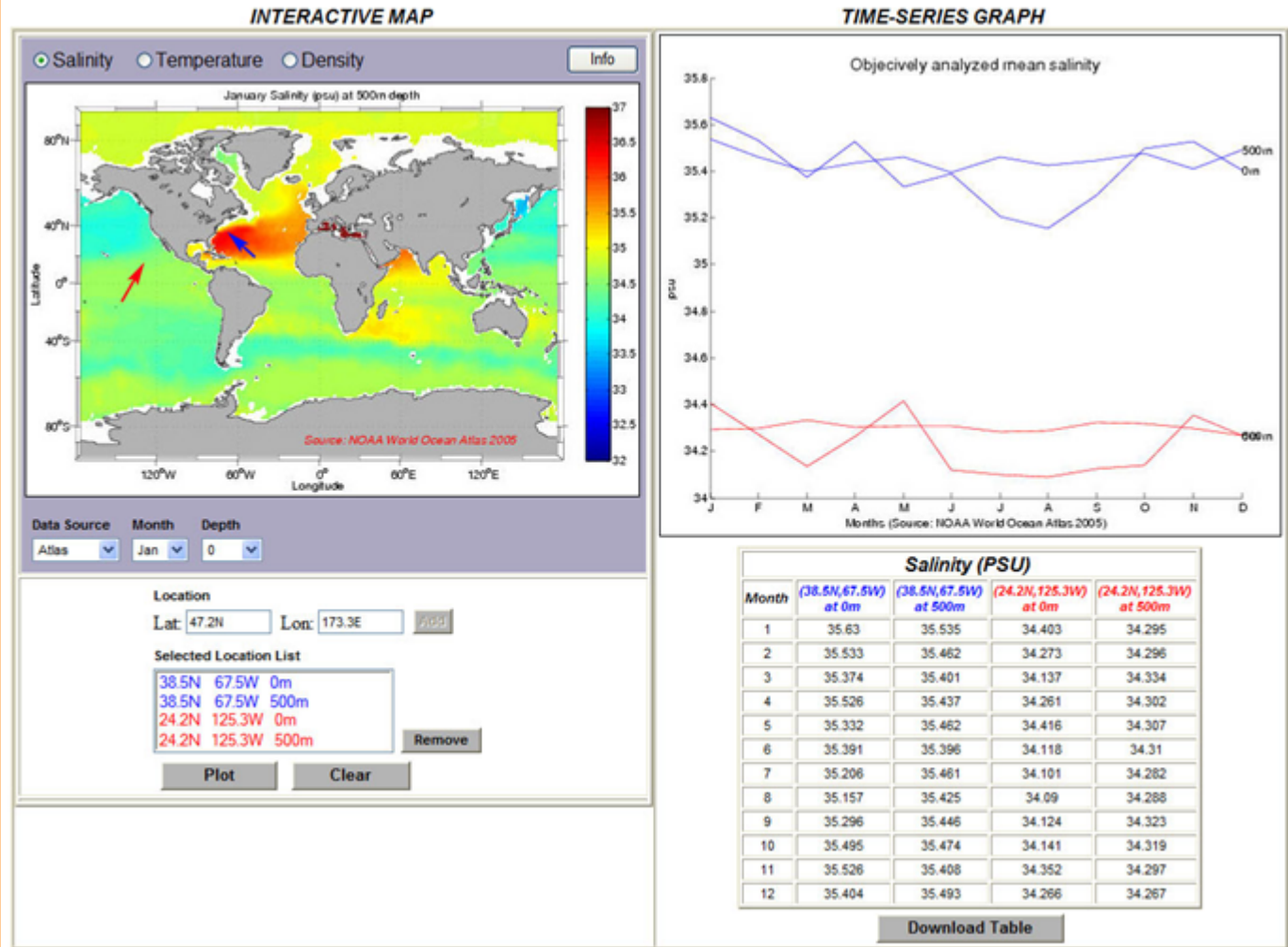
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